

**CHBE320 LECTURE IV  
MATHEMATICAL MODELING OF  
CHEMICAL PROCESS**

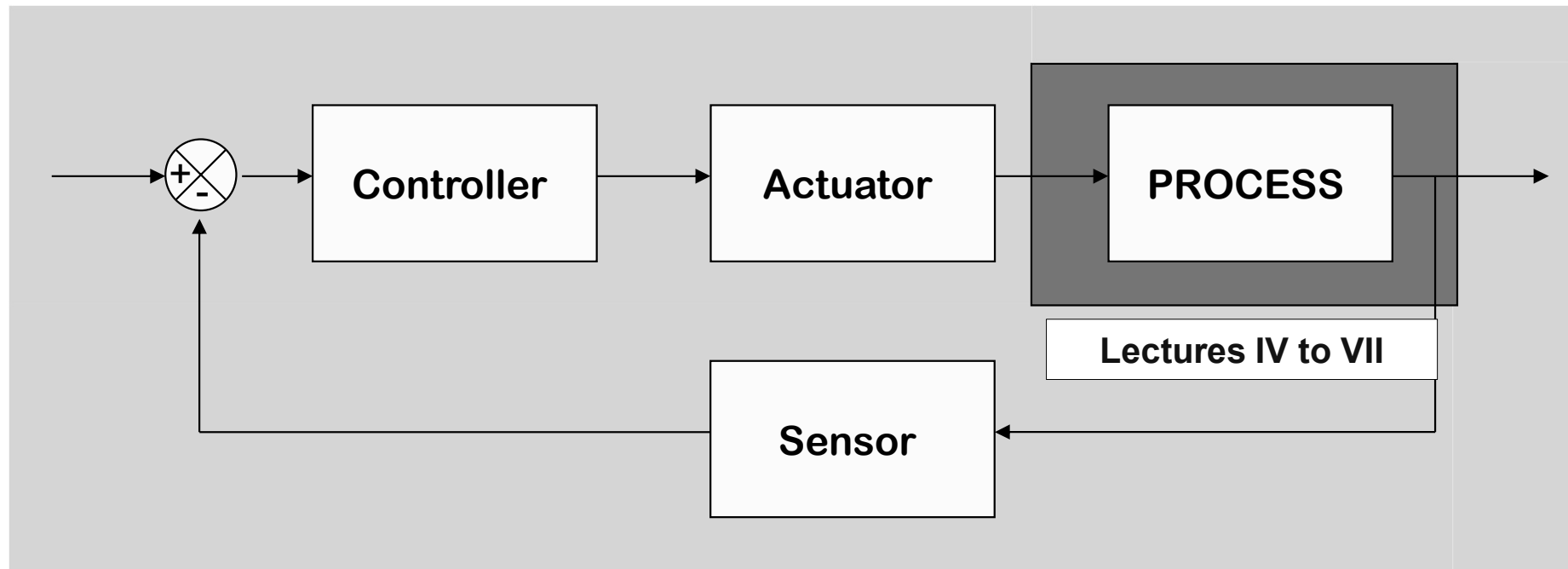
**Professor Dae Ryook Yang**

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**Dept. of Chemical and Biological Engineering  
Korea University**

# Road Map of the Lecture IV

- **Basics of Process Modeling**



- **Mathematical Modeling**
- **Steady-state model vs. Dynamic model**
- **Degree of freedom analysis**
- **Models of representative processes, etc.**

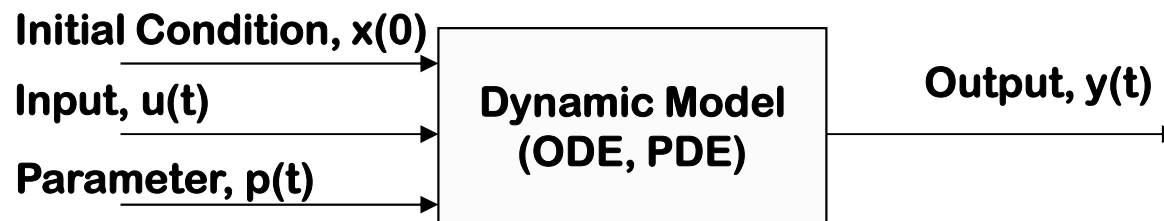
# THE RATIONALE FOR MATHEMATICAL MODELING

- **Where to use**
  - To improve understanding of the process
  - To train plant operating personnel
  - To design the control strategy for a new process
  - To select the controller setting
  - To design the control law
  - To optimize process operating conditions
- **A Classification of Models**
  - Theoretical models (based on physicochemical law)
  - Empirical models (based on process data analysis)
  - Semi-empirical models (combined approach)

# DYNAMIC VERSUS STEADY-STATE MODEL

- **Dynamic model**

- Describes time behavior of a process
  - Changes in input, disturbance, parameters, initial condition, etc.
- Described by a set of differential equations
  - : ordinary (ODE), partial (PDE), differential-algebraic(DAE)



- **Steady-state model**

- Steady state: No further changes in all variables
- No dependency in time: No transient behavior
- Can be obtained by setting the time derivative term zero

# MODELING PRINCIPLES

- **Conservation law**

- **Within a defined system boundary (control volume)**

$$\begin{aligned} \left[ \begin{array}{c} \text{rate of} \\ \text{accumulation} \end{array} \right] &= \left[ \begin{array}{c} \text{rate of} \\ \text{input} \end{array} \right] - \left[ \begin{array}{c} \text{rate of} \\ \text{output} \end{array} \right] \\ &+ \left[ \begin{array}{c} \text{rate of} \\ \text{generation} \end{array} \right] - \left[ \begin{array}{c} \text{rate of} \\ \text{disappearance} \end{array} \right] \end{aligned}$$

- **Mass balance (overall, components)**
- **Energy balance**
- **Momentum or force balance**
- **Algebraic equations: relationships between variables and parameters**

# MODELING APPROACHES

- **Theoretical Model**
  - Follow conservation laws
  - Based on physicochemical laws
  - Variables and parameters have physical meaning
  - Difficult to develop
  - Can become quite complex
  - Extrapolation is valid unless the physicochemical laws are invalid
  - Used for optimization and rigorous prediction of the process behavior
- **Empirical model**
  - Based on the operation data
  - Parameters may not have physical meaning
  - Easy to develop
  - Usually quite simple
  - Requires well designed experimental data
  - The behavior is correct only around the experimental condition
  - Extrapolation is usually invalid
  - Used for control design and simplified prediction model

# DEGREE OF FREEDOM (DOF) ANALYSIS

- **DOF**

- **Number of variables that can be specified independently**
- $N_F = N_V - N_E$ 
  - $N_F$  : **Degree of freedom (no. of independent variables)**
  - $N_V$  : **Number of variables**
  - $N_E$  : **Number of equations (no. of dependent variables)**
  - **Assume no equation can be obtained by a combination of other equations**

- **Solution depending on DOF**

- **If  $N_F = 0$ , the system is *exactly determined*. Unique solution exists.**
- **If  $N_F > 0$ , the system is *underdetermined*. Infinitely many solutions exist.**
- **If  $N_F < 0$ , the system is *overdetermined*. No solutions exist.**

# LINEAR VERSUS NONLINEAR MODELS

- **Superposition principle**

$\forall \alpha, \beta \in \mathfrak{R}$ , and for a linear operator,  $L$

Then  $L(\alpha x_1(t) + \beta x_2(t)) = \alpha L(x_1(t)) + \beta L(x_2(t))$

- **Linear dynamic model: superposition principle holds**

$\forall \alpha, \beta \in \mathfrak{R}, u_1(t) \rightarrow y_1(t)$  and  $u_2(t) \rightarrow y_2(t)$

$\alpha u_1(t) + \beta u_2(t) \rightarrow \alpha y_1(t) + \beta y_2(t)$

$\forall \alpha, \beta \in \mathfrak{R}, x_1(0) \rightarrow y_1(t)$  and  $x_2(0) \rightarrow y_2(t)$

$\alpha x_1(0) + \beta x_2(0) \rightarrow \alpha y_1(t) + \beta y_2(t)$

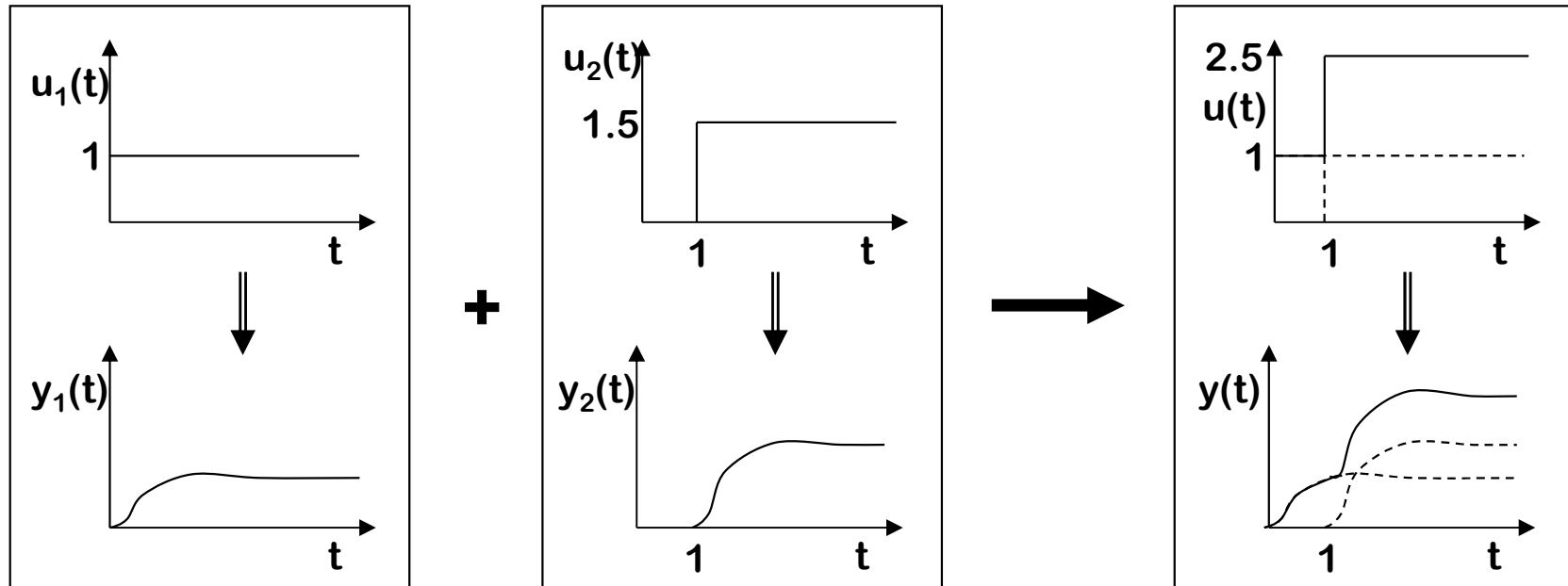
- **Easy to solve and analytical solution exists.**
- **Usually, locally valid around the operating condition**

- **Nonlinear: “Not linear”**

- **Usually, hard to solve and analytical solution does not exist.**



# ILLUSTRATION OF SUPERPOSITION PRINCIPLE



- **Valid only for linear process**

- For example, if  $y(t)=u(t)^2$ ,

- $(u_1(t)+1.5u_2(t))^2$  is not same as  $u_1(t)^2 +1.5u_2(t)^2$ .

# TYPICAL LINEAR DYNAMIC MODEL

- **Linear ODE**

$$\tau \frac{dy(t)}{dt} = -y(t) + Ku(t) \quad (\tau \text{ and } K \text{ are constant, 1st order})$$

$$\begin{aligned} \frac{d^n y(t)}{dt^n} + a_{n-1} \frac{d^{n-1} y(t)}{dt^{n-1}} + \dots + a_0 y(t) \\ = b_m \frac{d^m u(t)}{dt^m} + b_{m-1} \frac{d^{m-1} u(t)}{dt^{m-1}} + \dots + b_0 u(t) \quad (\text{nth order}) \end{aligned}$$

- **Nonlinear ODE**

$$\tau \frac{dy(t)}{dt} = -y(t)^2 + Ku(t)$$

$$\tau \frac{dy(t)}{dt} y(t) = -y(t) \sin(y) + Ku(t)$$

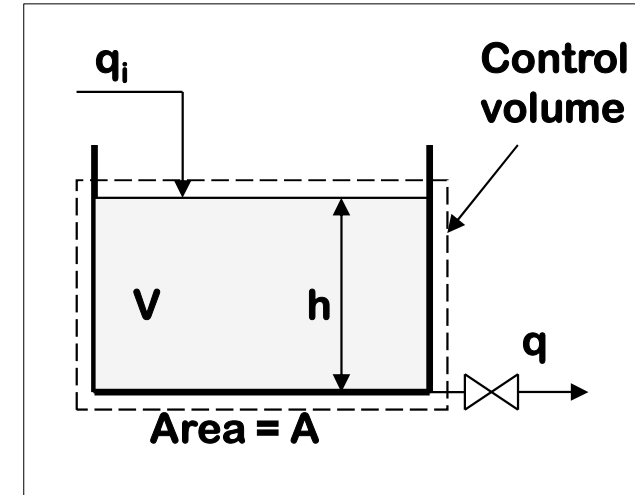
$$\tau \frac{dy(t)}{dt} = -y(t) + K\sqrt{u(t)}$$

$$\tau \frac{dy(t)}{dt} = -e^{-y(t)} + Ku(t)$$

# MODELS OF REPRESENTATIVE PROCESSES

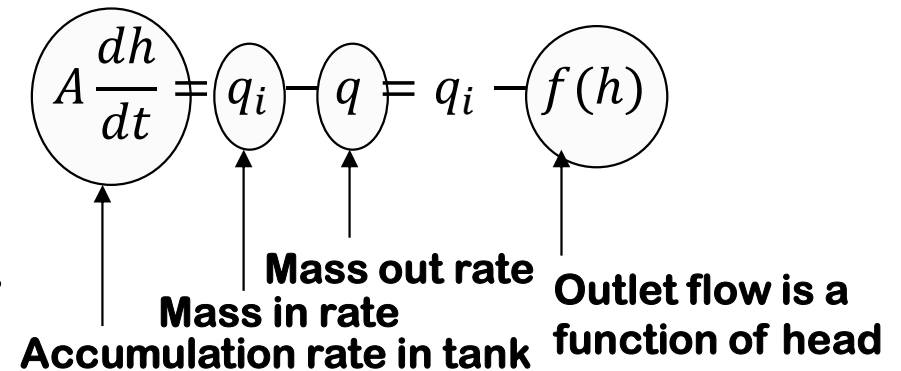
- **Liquid storage systems**

- System boundary: storage tank
- Mass in:  $q_i$  (vol. flow, indep. var)
- Mass out:  $q$  (vol, flow, dep. var)
- No generation or disappearance (no reaction or leakage)



- No energy balance

- **DOF=2** ( $h, q_i$ ) - 1=1
- If  $f(h) = h/R_V$ , the ODE is linear. ( $R_V$  is the resistance to flow)



- If  $f(h) = C_V \sqrt{\rho g h / g_c}$ , the ODE is nonlinear. ( $C_V$  is the valve constant)

- **Continuous Stirred Tank Reactor (CSTR)**

- Liquid level is constant (No acc. in tank)
- Constant density, perfect mixing
- Reaction:  $A \rightarrow B$  ( $r = k_0 \exp(-E/RT)c_A$ )
- System boundary: CSTR tank
- Component mass balance

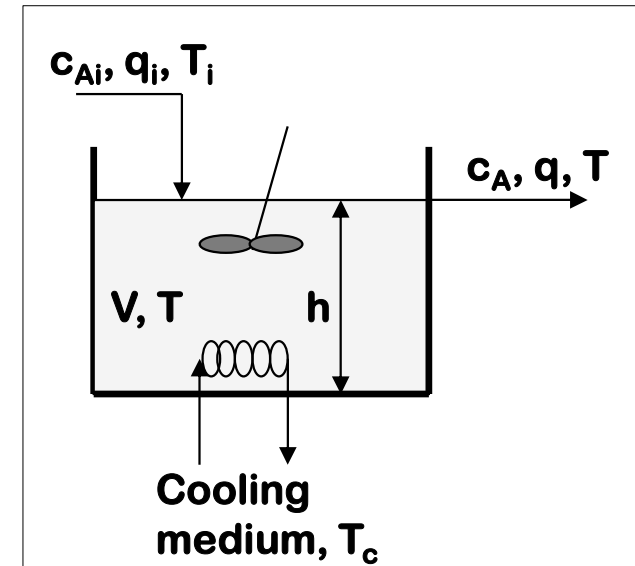
$$V \frac{dc_A}{dt} = q(c_{Ai} - c_A) - Vkc_A$$

- Energy balance

$$V\rho C_p \frac{dT}{dt} = q\rho C_p(T_i - T) + (-\Delta H)Vkc_A + UA(T_c - T)$$

- **DOF analysis**

- **No. of variables: 6** ( $q, c_A, c_{Ai}, T_i, T, T_c$ )
- **No. of equation: 2** (two dependent vars.:  $c_A, T$ )
- **DOF=6 - 2 = 4**
- **Independent variables: 4** ( $q, c_{Ai}, T_i, T_c$ )
- **Parameters: kinetic parameters,  $V, U, A$ , other physical properties**
- **Disturbances: any of  $q, c_{Ai}, T_i, T_c$ , which are not manipulatable**



# STANDARD FORM OF MODELS

From the previous example

$$\frac{dc_A}{dt} = \frac{q}{V}(c_{Ai} - c_A) - kc_A = f_1(c_A, T, q, c_{Ai})$$

$$\frac{dT}{dt} = \frac{q}{V}(T_i - T) + \frac{q}{\rho C_p}(-\Delta H)kc_A + \frac{UA}{\rho C_p}(T_c - T) = f_2(c_A, T, q, T_c, T_i)$$

- **State-space model**

$$\dot{\mathbf{x}} = d\mathbf{x}/dt = \mathbf{f}(\mathbf{x}, \mathbf{u}, \mathbf{d})$$

$$\text{where } \mathbf{x} = [x_1, \dots, x_n]^T, \mathbf{u} = [u_1, \dots, u_m]^T, \mathbf{d} = [d_1, \dots, d_l]^T$$

- **x: states**,  $[c_A \ T]^T$
- **u: inputs**,  $[q \ T_c]^T$
- **d: disturbances**,  $[c_{Ai} \ T_i]^T$
- **y: outputs** – can be a function of above,  $\mathbf{y}=\mathbf{g}(\mathbf{x},\mathbf{d},\mathbf{u})$ ,  $[c_A \ T]^T$
- **If higher order derivatives exist, convert them to 1<sup>st</sup> order.**

# CONVERT TO 1<sup>ST</sup>-ORDER ODE

- **Higher order ODE**

$$\frac{d^n x(t)}{dt^n} + a_{n-1} \frac{d^{n-1} x(t)}{dt^{n-1}} + \dots + a_0 x(t) = b_0 u(t)$$

- **Define new states**

$$x_1 = x, x_2 = \dot{x}, x_3 = \ddot{x}, \dots, x_n = x^{(n-1)}$$

- **A set of 1<sup>st</sup>-order ODE's**

$$\dot{x}_1 = x_2$$

$$\dot{x}_2 = x_3$$

⋮

$$\dot{x}_n = -a_{n-1}x_n - a_{n-2}x_{n-1} - \dots - a_0x_1 + b_0u$$

# SOLUTION OF MODELS

- **ODE (state-space model)**

- **Linear case: find the analytical solution via Laplace transform, or other methods.**
- **Nonlinear case: analytical solution usually does not exist.**
  - **Use a numerical integration, such as *RK method*, by defining initial condition, time behavior of input/disturbance**
  - **Linearize around the operating condition and find the analytical solution**

- **PDE**

- **Convert to ODE by discretization of spatial variables using *finite difference approximation* and etc.**

$$\frac{\partial T_L}{\partial t} = -v \frac{\partial T_L}{\partial z} + \frac{1}{\tau_{HL}} (T_w - T_L) \longrightarrow \frac{dT_L(j)}{dt} = -\frac{v}{\Delta z} T_L(j-1) - \left( \frac{v}{\Delta z} + \frac{1}{\tau_{HL}} \right) T_L(j) + \frac{1}{\tau_{HL}} T_w \quad (j = 1, \dots, N)$$
$$\frac{\partial T_L}{\partial z} \approx \frac{T_L(j) - T_L(j-1)}{\Delta z}$$

# LINEARIZATION

- **Equilibrium (Steady state)**

- Set the derivatives as zero:  $0 = f(\bar{x}, \bar{u}, \bar{d})$
- Overbar denotes the steady-state value and  $(\bar{x}, \bar{u}, \bar{d})$  is the equilibrium point. (could be multiple)
- Solve them analytically or numerically using Newton method, etc.

- **Linearization around equilibrium point**

- Taylor series expansion to 1<sup>st</sup> order

$$f(\mathbf{x}, \mathbf{u}) = f(\bar{\mathbf{x}}, \bar{\mathbf{u}}) + \frac{\partial f}{\partial \mathbf{x}} \bigg|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} (\mathbf{x} - \bar{\mathbf{x}}) + \frac{\partial f}{\partial \mathbf{u}} \bigg|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} (\mathbf{u} - \bar{\mathbf{u}}) + \dots$$

- Ignore higher order terms
- Define deviation variables:  $\mathbf{x}' = \mathbf{x} - \bar{\mathbf{x}}, \mathbf{u}' = \mathbf{u} - \bar{\mathbf{u}}$

$$\dot{\mathbf{x}}' = \frac{\partial f}{\partial \mathbf{x}} \bigg|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} \mathbf{x}' + \frac{\partial f}{\partial \mathbf{u}} \bigg|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} \mathbf{u}' = \mathbf{A}\mathbf{x}' + \mathbf{B}\mathbf{u}'$$

		Jacobian		
$\frac{\partial f}{\partial \mathbf{x}} =$	$\left[ \begin{array}{ccc} \frac{\partial f_1}{\partial x_1} & \dots & \frac{\partial f_1}{\partial x_n} \\ \vdots & \ddots & \vdots \\ \frac{\partial f_n}{\partial x_1} & \dots & \frac{\partial f_n}{\partial x_n} \end{array} \right]$			