

# Rapid synthesis of MOF-5 via microwave radiation

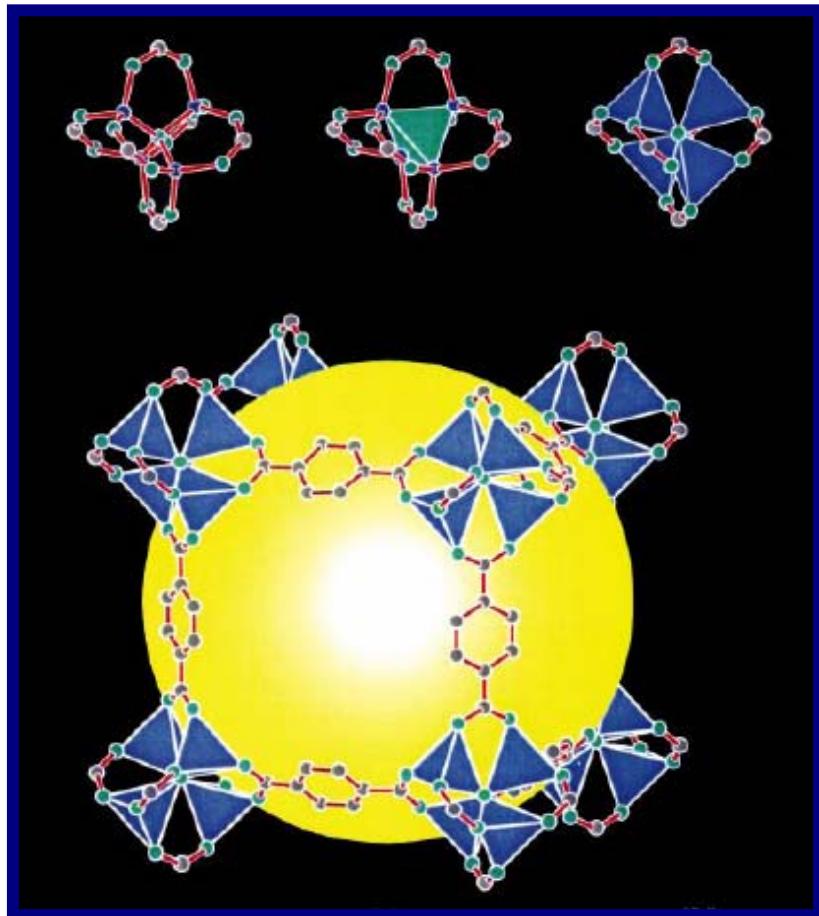


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# MOF(Metal Organic Framework)

## ● What is MOF(Metal Organic Framework)?



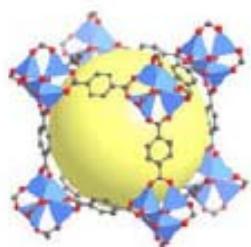
- 전이금속 이온이나 클러스터를 구조물의 꼭지로 하여 이들을 배위결합이 가능한 다양한 유기 리간드로 연결하여 높은 구조체.
- 다공성 화합물로 작용 가능. 현존하는 가장 비표면적이 큰 물질 중 하나.  
( $S_A > 5,000 \text{ m}^2/\text{g}$  가능.)
- 다양한 유기 리간드와 금속 이온의 사용으로 기공의 구조와 화학적 물성 조절 가능.

대표적 MOF인 MOF-5  $[\text{Zn}_4(\text{O})\text{O}_{12}\text{C}_6]$

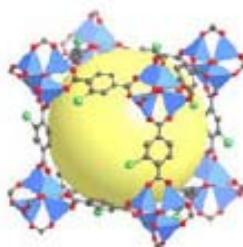
# MOF (Metal Organic Framework)

## ● Isoreticular Metal-Organic Frameworks (IRMOFs)

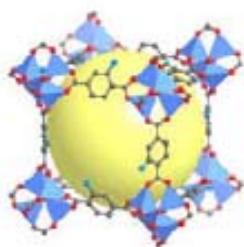
a series of materials having the same underlying topology and (typically) constructed from the same secondary building units



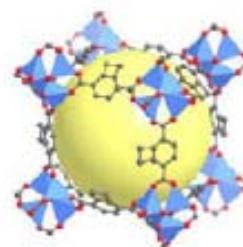
IRMOF-1



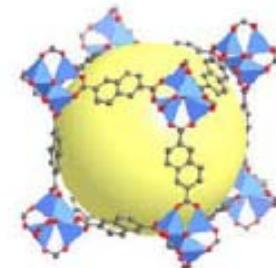
IRMOF-2



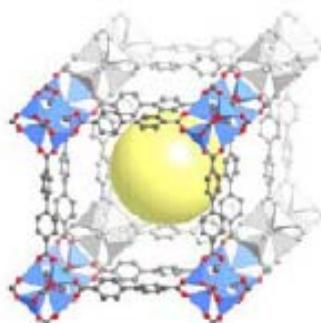
IRMOF-3



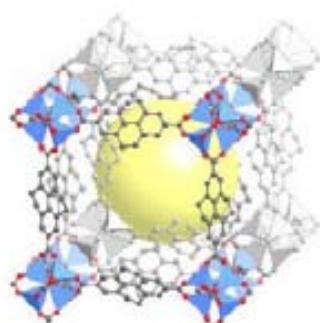
IRMOF-6



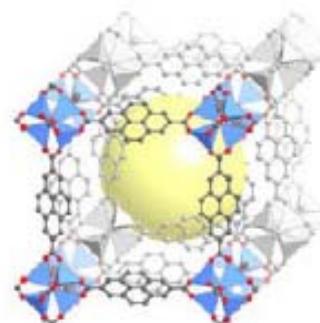
IRMOF-8



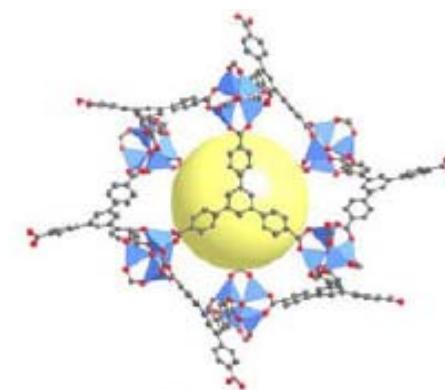
IRMOF-9



IRMOF-11



IRMOF-13



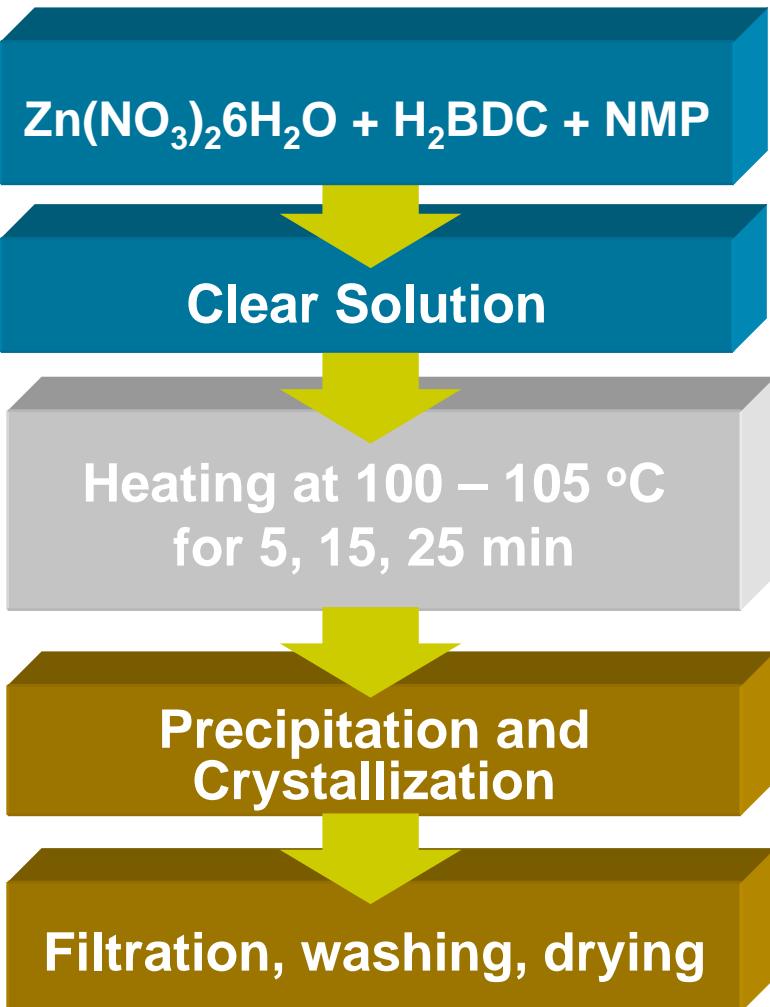
MOF-177

M. Eddaoudi, J. Kim, N. Rosi, D. Vodak, J. Wachter, M. O'Keeffe, O.M. Yaghi, *Science*, **2002**, 295, 469.

H. K. Chae, D. Y. Siberio-Perez, J. Kim, Y-B. Go, M. Eddaoudi, A. J. Matzger, M. O'Keeffe, O.M. Yaghi, *Nature*, **2004**, 427, 523.

# EXPERIMENTAL

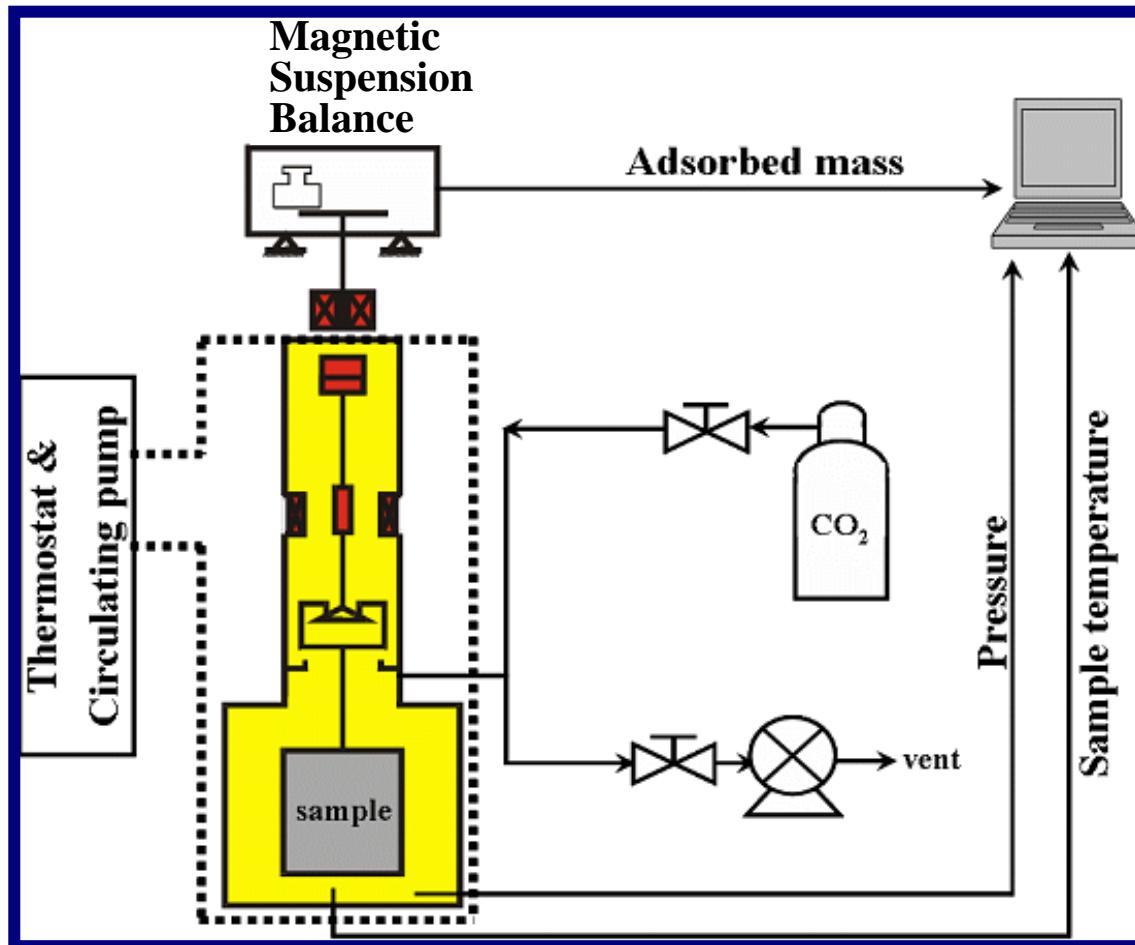
## ● Synthesis flow sheet



Microwave synthetic system

# Equipment

## ● CO<sub>2</sub> 흡착/저장 성능의 정량화 : gravimetric analysis



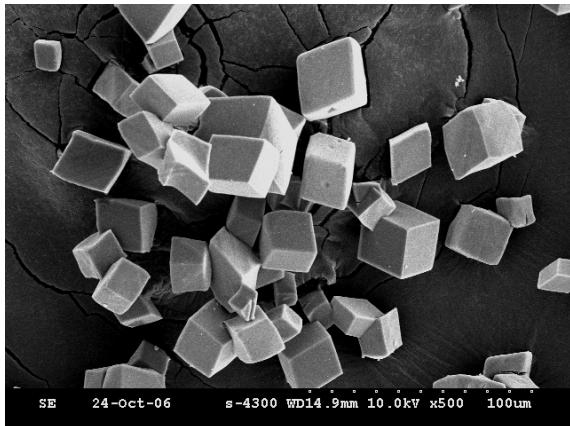
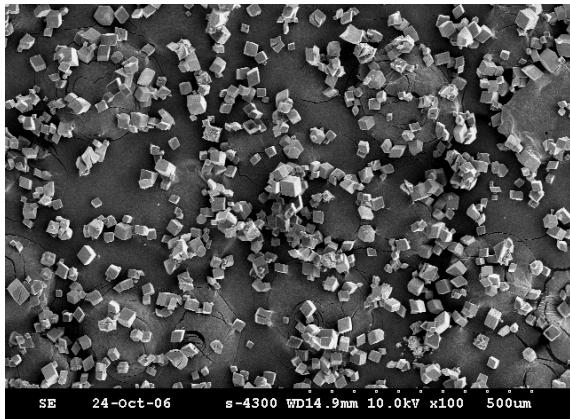
- 흡착등온선 (1-40 기압, 상온)
- 흡탈착 재생성 (Pressure Swing sorption).
- TGA 기능을 이용한 흡착제의 재생성 분석 (Temperature Swing Analysis).
- 흡/탈착 속도 측정.

Rubotherm  
Magnetic Suspension Balance  
- Resolution :  $10^{-5}$  g  
- Max. Weight : 80 g

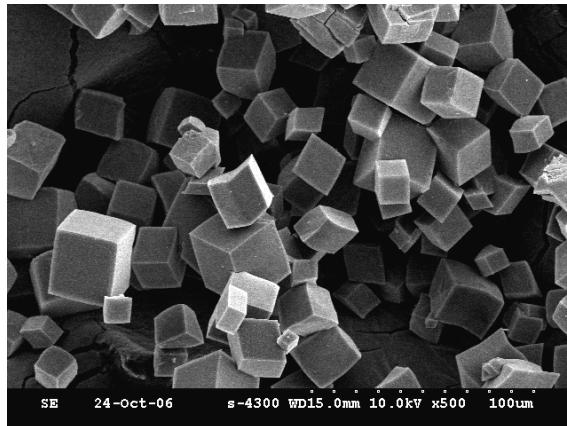
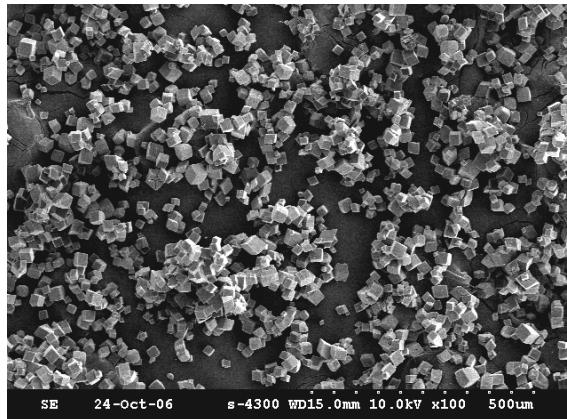
# RESULTS & DISCUSSION

## ● Scanning Electron Microscope image

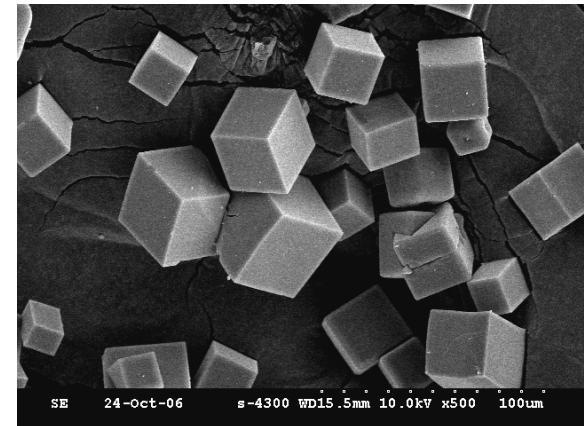
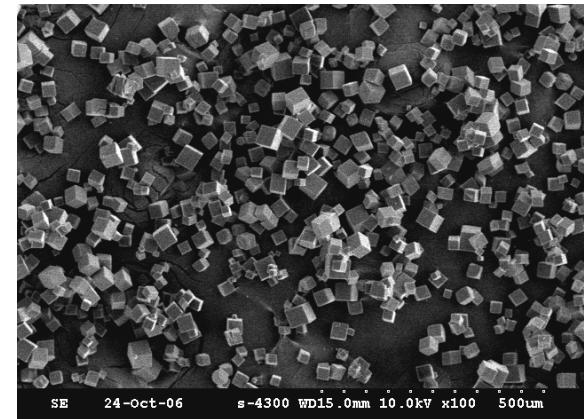
Power 25%, 5 min



Power 25%, 15 min



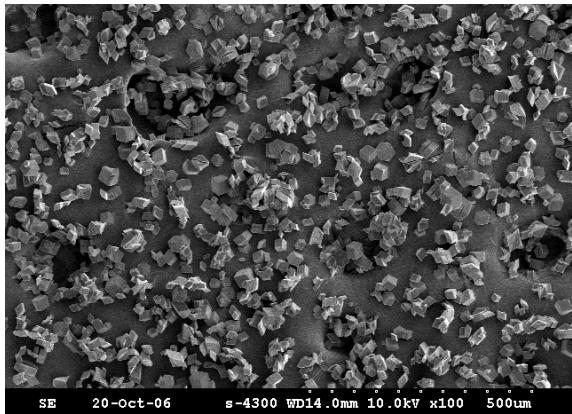
Power 25%, 25 min



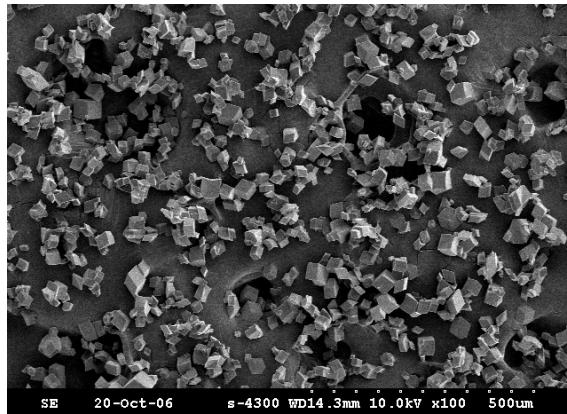
# RESULTS & DISCUSSION

## ● Scanning Electron Microscope image

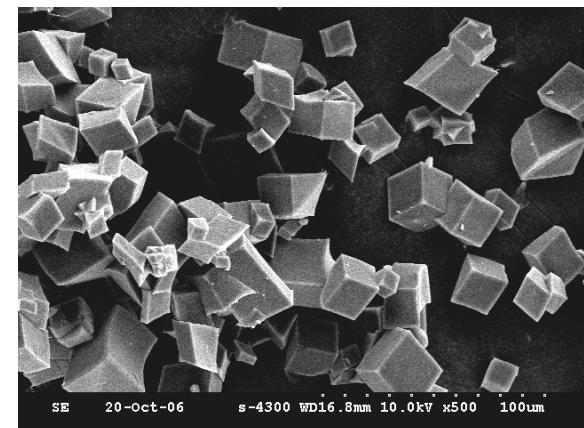
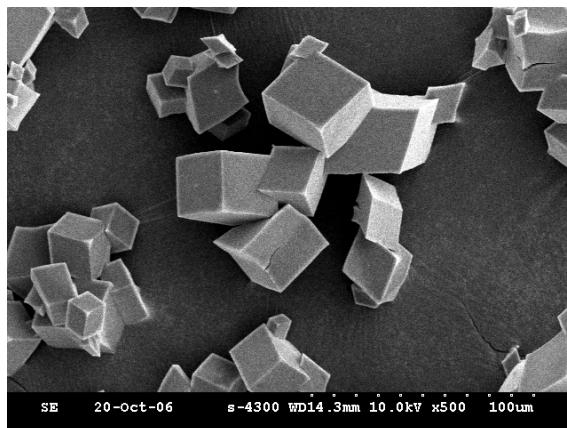
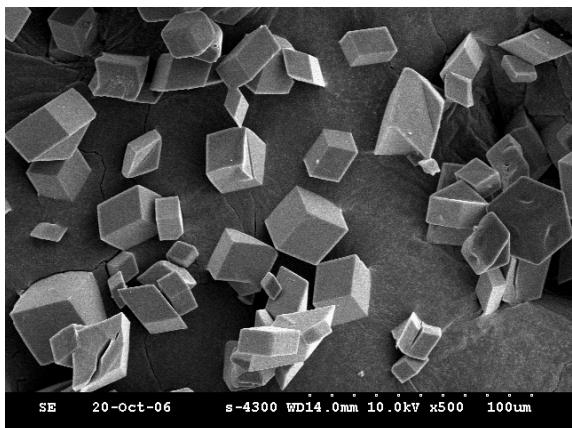
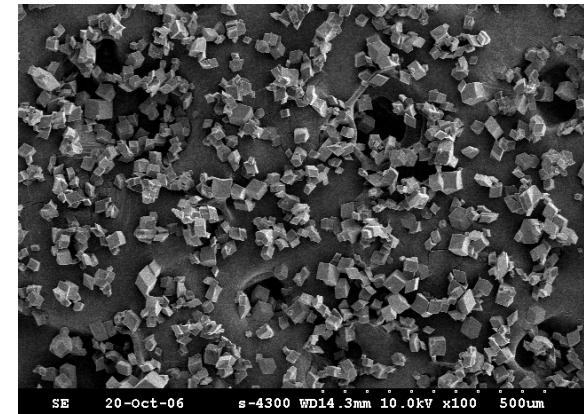
Power 75%, 5 min



Power 75%, 15 min

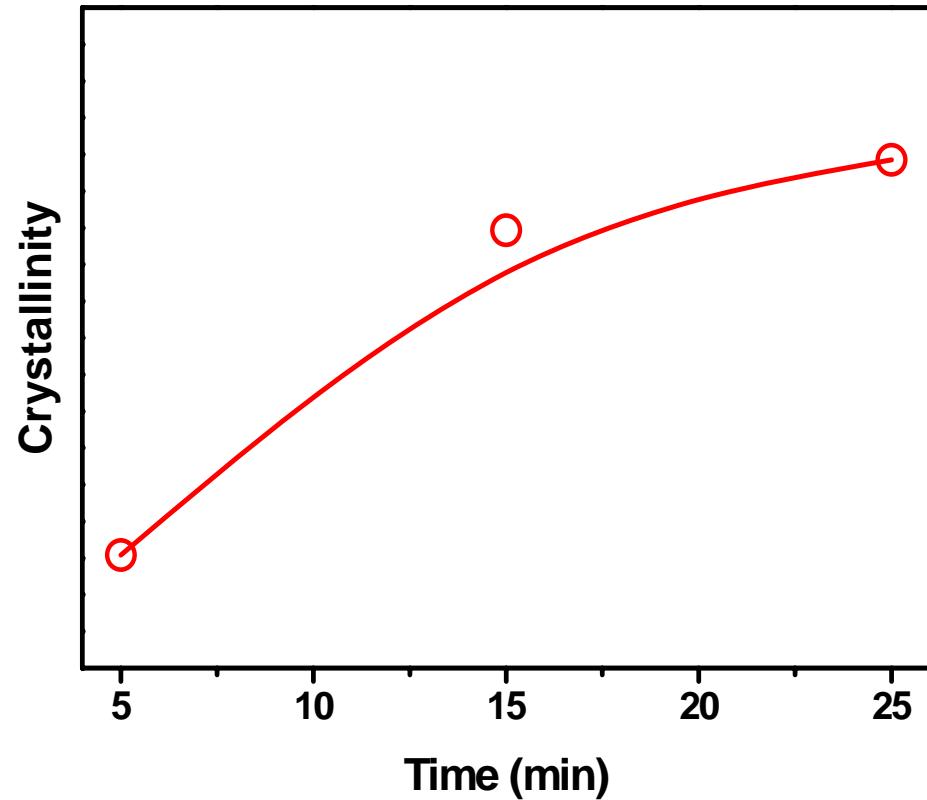
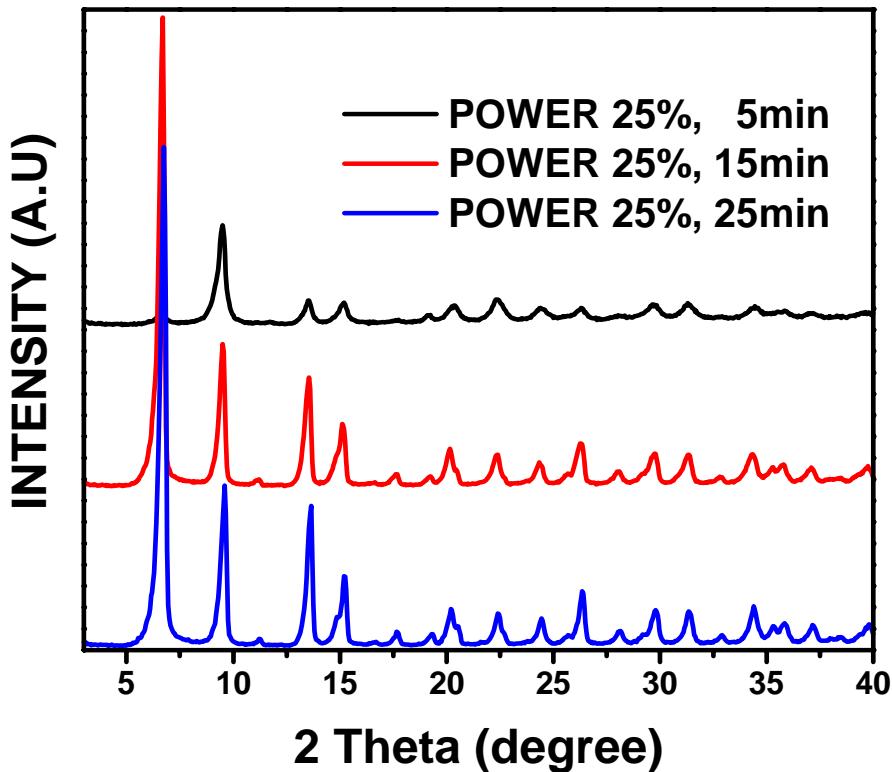


Power 75%, 25 min



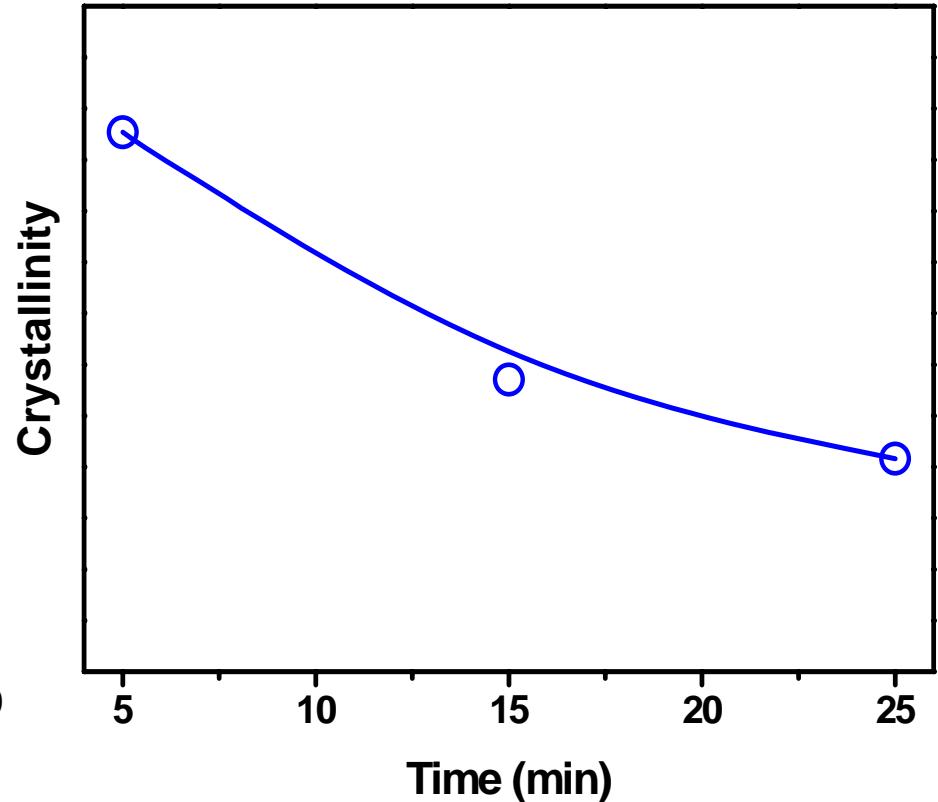
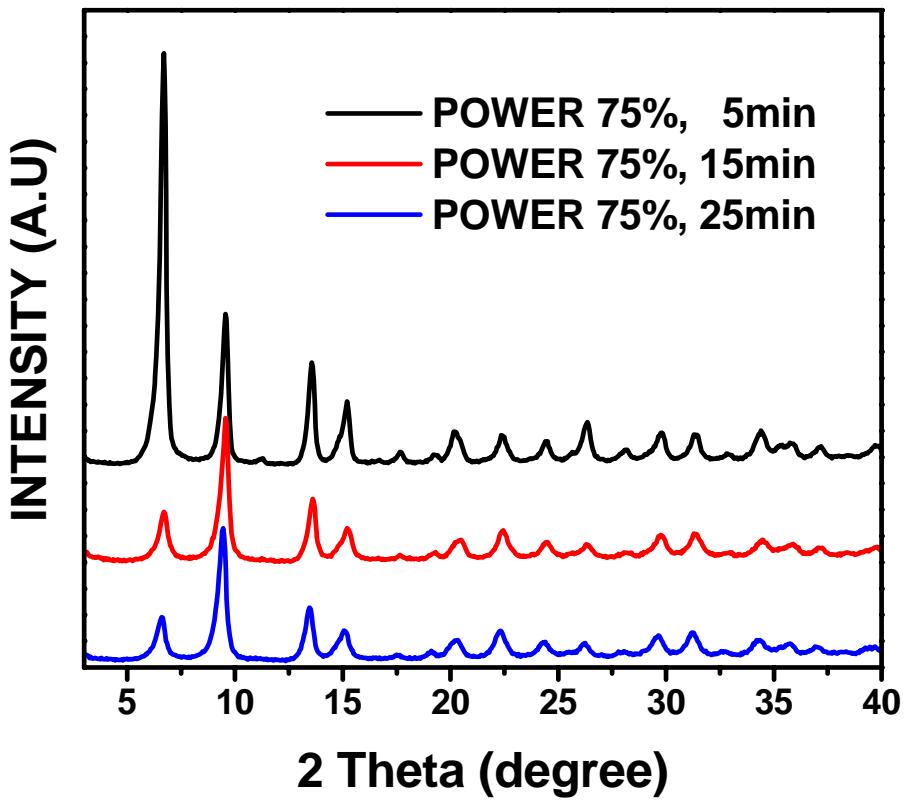
# RESULTS & DISCUSSION

## ● X-Ray diffraction patterns



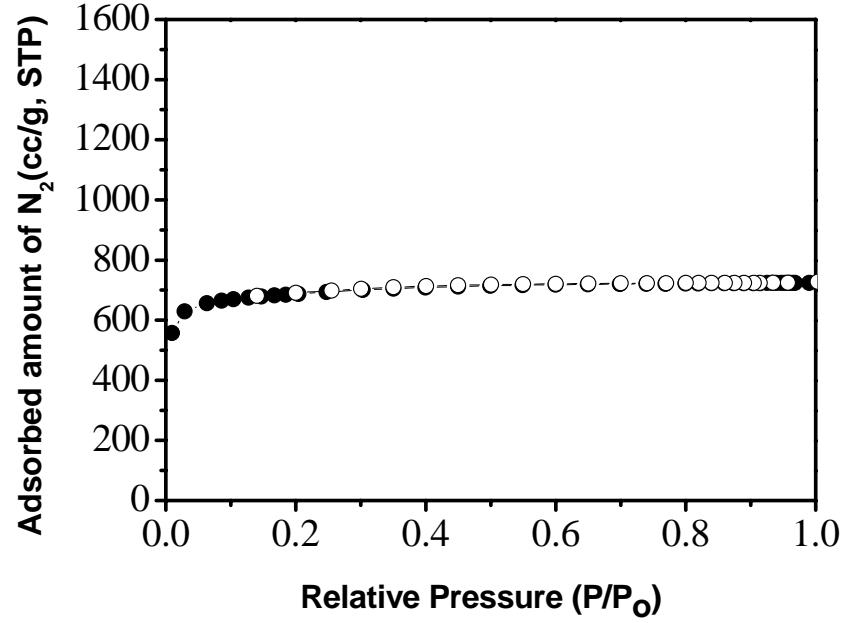
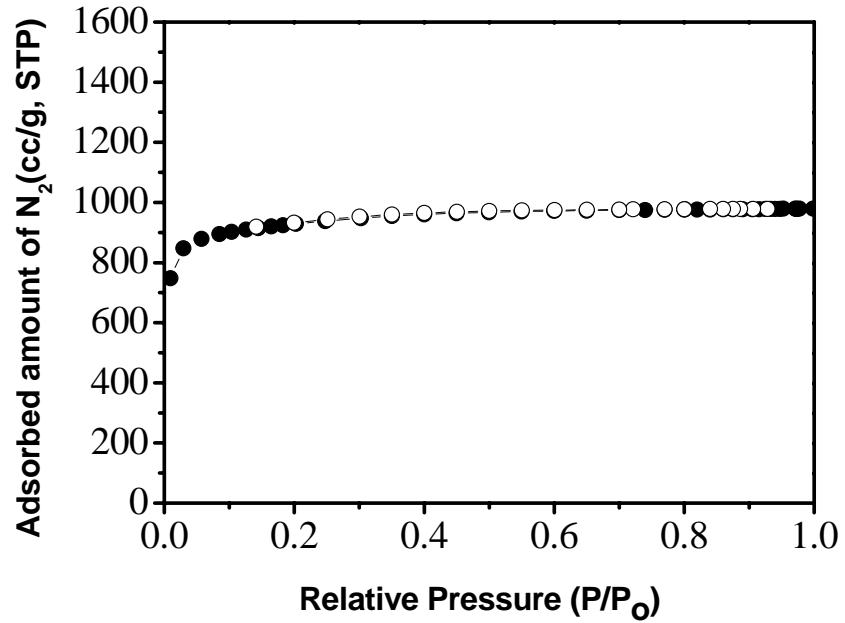
# RESULTS & DISCUSSION

## ● X-Ray diffraction patterns



# RESULTS & DISCUSSION

## ● Nitrogen Ad/Desorption analysis



Power 25%

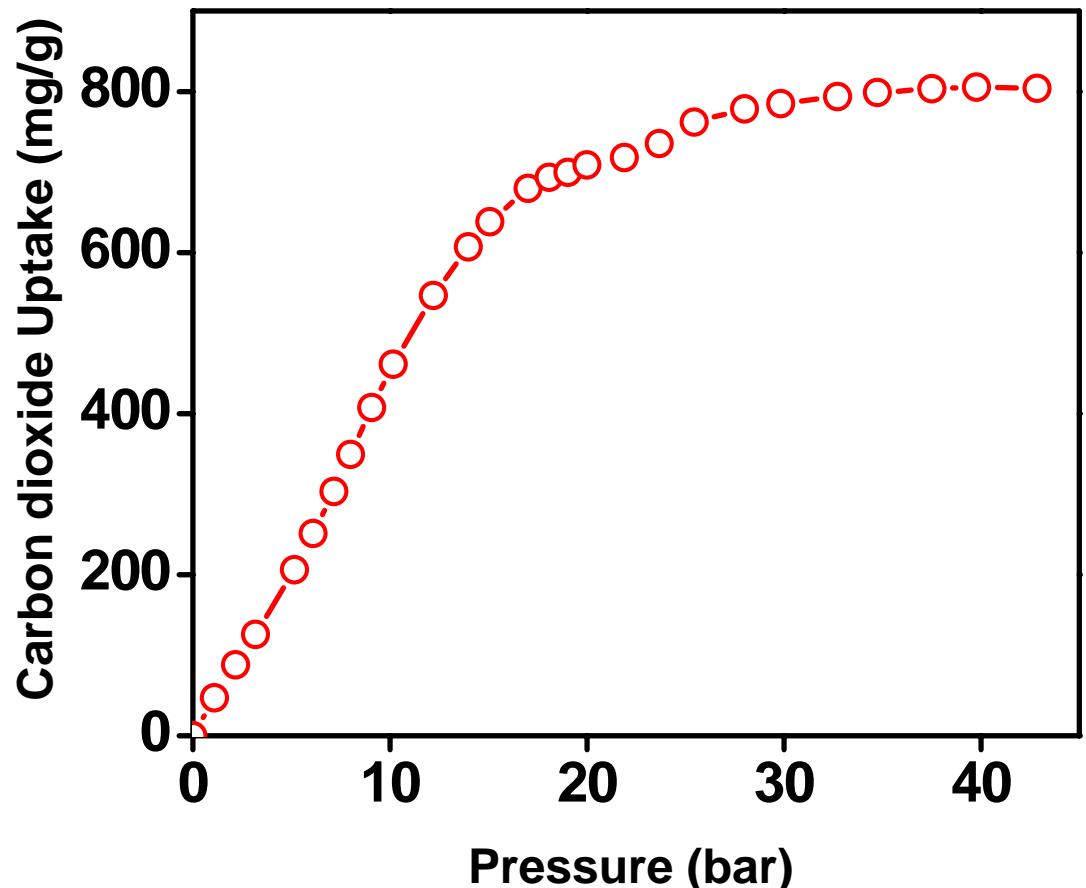
Langmuir Area ( $\text{m}^2/\text{g}$ )	3200–4000
Micropore Area ( $\text{m}^2/\text{g}$ )	2400–3000
Pore volume (cc/g)	1.20–1.60
Micropore Volume (cc/g)	1.13–1.45

Power 75%

Langmuir Area ( $\text{m}^2/\text{g}$ )	2700–3000
Micropore Area ( $\text{m}^2/\text{g}$ )	2000–2200
Pore volume (cc/g)	1.00–1.12
Micropore Volume (cc/g)	0.95–1.07

# RESULTS & DISCUSSION

## ● CO<sub>2</sub> adsorption capacity of microwave synthesized MOF-5



# Conclusions

- MOF-5 was successfully synthesized by microwave heating with ca. 3000-4000 m<sup>2</sup>/g surface area.
- Synthesis period of MOF-5 was reduced from 2 days to 5-25 min.
- CO<sub>2</sub> adsorption capacity of MOF-5 synthesized by microwave was 805 mg/g at 40 bar.