## The effect of concentration of extractants and pH of solutions on extraction of Pd ions in Pd/Sn colloid solution

<u>여길환</u>, 홍원희\*, 전상준, 이홍기<sup>1</sup> 한국과학기술원; <sup>1</sup>한국생산기술연구원 (hongwh@kaist.ac.kr\*)

Noble metal like a Pd should be reused because of their rarity on earth and the price of it. Pd ions are used as a catalyst for electroless plating process. The mixed PdCl<sub>2</sub>/SnCl<sub>2</sub> solution(Pd/Sn colloid particle) containing hydrochloric acid is most commonly used in electroless plating to make a nonmetallic substrate surface catalytically active to initiate the metal deposition. PdCl<sub>2</sub>/SnCl<sub>2</sub> solution has a Pd-Sn alloy core with a stabilizing layer of adsorbed Sn(+2) ions. Sn(+2) ions reduce a Pd ion and make a stabilizing layer in the surface of Pd/Sn particle. Pd ions are extracted by Tri-n-octylphosphine oxide and Thenoyltrifluoroacetone. This study investigates the effect of concentration of extractants and pH of solutions on extraction of Pd ions.