Enzymatic polymerization of resorcinol alkylate and bisphenol A using CiP(*Coprinus cinereus* peroxidase) application to photoresist

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Resorcinol alkylate and bisphenol A were oxidatively polymerized by CiP peroxidase, which is expected to be an alternative for the preparation of conventional phenolic resins without using formaldehyde. Water/solvent ratio gave significant influences on the enzymatic polymerization of resorcinol alkylate and bisphenol A. To know the mechanism of enzyme deactivation that solvents and hydrogen peroxide induce is critically important to obtain higher yield of phenol polymer product. Hydrogen peroxide being used as oxidizing agent induced deactivation of peroxidase by destruction of heme structure. The stability of peroxidase was significantly improved by adding organic solvent. This is thought that radical scavengering by 2–propanol may play a major role in protecting peroxidase from the oxidation of oxidizing radical such as peroxyl and alkoxyl and superoxide anion radicals.