Synthesis of the lithium titanate(Li₄Ti₅O₁₂) particles in supercritical methanol

<u>이영호</u>^{1,2}, 안기호^{1,2}, 송봉근^{1,2}, 이윤우^{1,2,*} ¹서울대학교 공과대학 화학생물공학부; ²서울대학교 화학공정신기술연구소 (vwlee@snu.ac.kr*)

Spinel lithium titanate ($\text{Li}_4\text{Ti}_5\text{O}_{12}$) is widely used as an anode material in the lithium–ion batteries due to its small volume change during the intercalation/deintercalation of lithium ions and its long life–cycle. In conventional method, $\text{Li}_4\text{Ti}_5\text{O}_{12}$ particles have been prepared by the solid–state reaction, which requires the high reaction temperature (700 ~ 1000 °C) and long reaction time (10 ~ 24 hrs).

In this study, $\mathrm{Li_4Ti_5O_{12}}$ was prepared in the supercritical methanol and supercritical water. Li–Ti–O precursors were prepared by mixing the lithium hydroxide aqueous solution and 2–propanol containing titanium isopropoxide, and Li–Ti–O precursors were redispersed in water or methanol. The reaction was performed at 400 °C and 300 bar. The molar ratio of lithium hydroxide and titanium isopropoxide was varied from 4:5 to 2:1. In addition, the effect of reaction medium on the crystalline structure of the product was also investigated.