Fabrications and Eletrochemical analysis of WC derived from conventional TPR method

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Recently tungsten carbides have been considered as an alternating electrocatalyst at the anode and cathode electrode of low temperature fuel cells, because they show high activity for electro-oxidation of methanol, hydrogen and electro-reduction of oxygen with minimal loading of platinum. We have fabricated the various structures and phases of tungsten carbides preserved the high physical properties and good electro-activity via conventional modified TPR method by using WO3 and H2WO4. Especially, the mass activity (mA/mg of Pt taken at 0.75V – Ag/AgCl) of all 7.5wt% Pt loaded tungsten carbide based catalysts show higher activity for methanol oxidation than commercial 20wt% PtRu/C (E-Tek) catalyst.