## Structure and phase equilibrium of tert-butyl nitrite + CH<sub>4</sub> hydrate

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Gas hydrates has been considered as one of potential materials for gas storage and transportation. In particular, structure–H gas hydrate has been highlighted due to its higher storage capacity and thermal stability than structure–I or structure–II hydrate. In this study, we found a new structure–H former, *tert*–butyl nitrite, through identifying its CH<sub>4</sub> containing hydrate structure and guest distribution by powder X–ray diffraction and high–power decoupling <sup>13</sup>C NMR spectroscopy. The phase equilibrium of *tert*–butyl nitrite + CH<sub>4</sub> hydrate were also investigated and confirmed to be more stable than pure CH4 hydrate. With high polarity of *tert*–butyl nitrite expecting to improve hydrate formation kinetics, the present findings might contribute to the application of gas hydrate to natural gas storage and transportation.