

Isobaric vapor–liquid equilibria for the binary system of n-Heptane and ethylene glycol mono-propyl ether(C3E1)

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This paper shows isobaric vapor–liquid equilibria (VLE) for the binary system of n-Heptane and Ethylene glycol monopropyl ether (C3E1). Alkoxyethanol such as C3E1 has associating groups so that it shows a highly non-ideal phase behavior due to the H-bond interaction. This isobaric VLE system experimented at different pressures for the measurement of the associating fluid mixture by Fischer VLE 602. The correlation for the experimented data uses the two term virial equation for vapor–phase fugacity coefficients and the three suffix Margules equation, Wilson, NRTL, UNIQUAC for liquid–phase activity coefficients. The results of the experiment shows a good agreement with variety of models.