

Efficient electrooxidation of methanol and ethanol using MoO_x-decorated Pd catalysts in alkaline media

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Methanol and ethanol may be used as portable clean sources of liquid fuels in AEM fuel cells. Pd-based catalysts are promising candidate catalysts for the efficient electrooxidation of alcohol. Effective approach has involved the addition of oxide materials to a Pd catalyst to improve the oxidation of small alcohols in alkaline media. Metal oxide can easily form hydroxyl species (OH⁻) on such metal oxide surfaces and can help the oxidation of adsorbed CO or intermediate species on a catalyst surface. In this study, we study the preparation of a MoO_x-decorated Pd catalyst supported on carbon for the electrooxidation of small alcohols in alkaline media. We compared the electrocatalytic activities of two catalyst materials upon modification with MoO_x: Pd nanoparticle-deposited MoO_x/C and MoO_x-deposited Pd/C. Among the materials, the Pd-MO/C catalyst demonstrated a much higher mass activity toward methanol or ethanol electrooxidation in alkaline media. This work was supported by the KETEP (no.20103020030020-11-2-200) and the Global Frontier R&D Program (no.0420-20110157).