

**Mesoporous cobalt-manganese oxide catalysts for preferential CO oxidation in H<sub>2</sub>-rich stream**

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Hydrogen, the fuel of PEMFC, is produced via the reforming of hydrocarbons and water-gas shift reaction were contains 0.5vol % - 2.0 vol % CO, which can poison Pt anode. Therefore, the concentration of CO must be reduced to below 100ppm. It is known that the preferential oxidation (PROX) of CO in H<sub>2</sub>-rich is the simplest and cost-effective routes for the removal of CO in H<sub>2</sub>. The catalysts reported for PROX include supported noble metal catalysts such as Pt, Rh, Ru and Au catalysts and transition metal oxide catalysts, such as Co<sub>3</sub>O<sub>4</sub>, MnO<sub>2</sub>, CuO-CeO<sub>2</sub>, Co<sub>3</sub>O<sub>4</sub>-CeO<sub>2</sub> has been recognized as one of the promising candidates. Among these the Co<sub>3</sub>O<sub>4</sub> was known to exhibit good catalytic performance for the oxidation of CO by O<sub>2</sub> at lower temperatures. Also manganese catalysts recognize as a promoter cobalt oxide and copper oxide catalysts. Adding MnO<sub>x</sub> can be improved catalysts dispersion and interaction between support and active metal.

In this present work, we were synthesized mesoporous Co<sub>3-x</sub>Mn<sub>x</sub>O<sub>4</sub> catalysts by using nano-replication method from mesoporous silica template of KIT-6. The mesoporous Co<sub>3-x</sub>Mn<sub>x</sub>O<sub>4</sub> catalysts exhibited high catalytic activity compared with both Co<sub>3</sub>O<sub>4</sub> and Mn<sub>2</sub>O<sub>3</sub>.