

### The H<sub>2</sub>S tolerance of ceria coated Ni/Al anode for molten carbonate fuel cells

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The hydrosulfid included in fuel gas is well established as an impurity for the molten carbonate fuel cells (MCFCs). H<sub>2</sub>S content is limited to 100 ppm in fuel gas. The Ni based anode of the MCFC reacts with H<sub>2</sub>S to form nickel sulfide, which can block active electrochemical reaction sites. And it is directly concerned in the cell performance. The ceria is a good candidate material as a desulfurization sorbent. It is well known that ceria reacts with H<sub>2</sub>S to form cerium oxysulfide and water. For this reason, we used ceria coated Ni-Al alloy anode to enhance the cell performance of anode in the MCFCs. The ceria coated anode was prepared by vacuum suction method using ceria sol. The physical properties of the modified anode were evaluated using the SEM, XRD, Hg porosimeter. To analyze the effect of sulfur on the cell performance, the MCFC single cell (10cm × 10 cm) was used. The MCFC single cell was operated at 650°C under a current density of 150mA/cm<sup>2</sup>. During the introduction of H<sub>2</sub>S, the I-V curve was measured under H<sub>2</sub>S concentration in the range of 100 ppm. These results were compared with the single cell performance used uncoated anode.