Synthesis and Characterization of Ether-functionalized Imidazolium-based Ionic Liquids and their SO_2 absorption Properties

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A series of ether–functionalized imidazolium–based ionic liquids was prepared and characterized using FT–IR, ^1H –NMR, TGA and DSC. All of them were liquid at room temperature and thermally stable up to 300°C. Among the prepared ionic liquids, bis (trifluoromethanesulfonyl) imide ([Tf $_2\text{N}^-$]) and nonafluorobutyl sulfonate ([C $_4\text{F}_9\text{SO}_3$] $^-$) showed hydrophobicity, while other anion such as dicyanamide ([(CN) $_2\text{N}$] $^-$) and trifluoromethane sulfonate ([CF $_3\text{SO}_3$] $^-$) showed hydrophilicity. Using the prepared ionic liquids, SO $_2$ absorption/desorption experiments were conducted and compared their SO $_2$ absorption capacity.