

Decomposition of H₂S on Ni(100) and Ni₃Al(100):
A first-principles study

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Spin-polarized density functional theory studies of H₂S decomposition on Ni(100) and Ni₃Al(100) surfaces were conducted to understand the aluminum(Al) alloying effect on the H₂S dissociation. We first determined the near surface structure of fully ordered Ni₃Al alloy along the [100] direction and examined the activation barriers for the H₂S and HS decomposition by using Climbing Image-Nudged Elastic Band method. We found that Al atom in bimetallic Ni₃Al(100) tends to exist in the first surface layer, rather than in the second or third layer, and the Ni₃Al (100) surface can substantially retard the H₂S decomposition by reducing the adsorption energy of sulfur compounds compared to the pure Ni(100) case. Furthermore, we found that the modification of surface Ni atoms by Ni-Al electronic (ligand) interaction and vertical location of Al in the Ni₃Al(100) slab, in particular, the alteration of the Ni *d* state induced by the first surface layer Al (rather than the second layer Al), serves an important role in reducing the adsorption energy of sulfur compounds.