

Preparation and characterization of ZnFe<sub>2</sub>O<sub>4</sub> heterostructures as ultra-high sensitive electrode for formaldehyde chemical sensor

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Zinc ferrite (ZnFe<sub>2</sub>O<sub>4</sub>) heterostructures were synthesized through low temperature solvothermal process. The bulk electrode of ZnFe<sub>2</sub>O<sub>4</sub> was prepared in the form of pellet and coated with graphene for the effective detection of toxic formaldehyde chemical. The phase purity and composition of ZnFe<sub>2</sub>O<sub>4</sub> were confirmed by elemental analysis such as Raman Spectroscopy and X-Ray diffraction studies. From room temperature photoluminescence (PL) spectroscopy, the synthesized ZnFe<sub>2</sub>O<sub>4</sub> exhibited prominent band-edge ultraviolet (UV) and weak visible emission (violet), indicated the optical properties of ZnFe<sub>2</sub>O<sub>4</sub>. The electrochemical revealed that the prepared ZnFe<sub>2</sub>O<sub>4</sub> heterostructures based electrode showed the sensing response toward formaldehyde. The enhanced electrochemical performances could be attributed to the synergistic effect between ZnFe<sub>2</sub>O<sub>4</sub> and graphene and the observed results demonstrated ZnFe<sub>2</sub>O<sub>4</sub> heterostructures as promising probe candidate for the fabrication of highly sensitive and selective chemical sensors.