

CO Oxidation on SnO₂ Surfaces Promoted by Pt-Doping

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Doping a precious metal atoms into host metal oxide can enhance catalytic activity by changing the property of surface lattice oxygen. Here, a Pt-doped antimony-tin oxide (Pt/Sb-SnO₂) for CO and C₃H₆ oxidation was synthesized. Pt-deposited tin oxide (Pt/SnO₂) and silica (Pt/SiO₂) were also prepared for comparison. HAADF-STEM images, DRIFT for CO chemisorption, H₂-uptakes, and XPS results showed that the Pt/Sb-SnO₂ has atomically doped Pt inside the tin oxide surface. The Pt/Sb-SnO₂ showed the highest activity for CO oxidation but the poorest activity for C₃H₆ oxidation. CO-TPR and O₂-TPD results showed that doping Pt atoms into the tin oxide lattice leads to a better surface lattice oxygen activity, resulting in enhanced catalytic performance. In-situ DRIFT for CO oxidation showed less formation of carbonates on the surface of the Pt/Sb-SnO₂, resulting in enhanced CO oxidation activity and durability.