The effect of co-precipitation pH on physico-chemical properties of $CuO-ZnO-Al_2O_3$ catalyst for the low temperature water-gas shift reaction

<u>안선용</u>, 나현석, 심재오, 장원준, 전경원, 김학민, 이열림, 유성연, 김경진, 김범준, 조재완, 노현석[†] 연세대학교 (hsroh@yonsei.ac.kr[†])

The low temperature water-gas shift (LT-WGS) reaction has been carried out at gas hourly space velocity (GHSV) of 8,001 h $^{-1}$ over CuO-ZnO-Al $_2$ O $_3$ catalyst. To investigate the effect of co-precipitation pH on physico-chemical properties of CuO-ZnO-Al $_2$ O $_3$ catalyst for the low temperature water-gas shift reaction, pH values of co-precipitation solution were systemically changed. The effect of pH on co-precipitated CuO-ZnO-Al $_2$ O $_3$ catalyst has been interpreted through characterization of BET, TPR, XRD, N $_2$ O chemisorption and related to catalytic activity results in the LT-WGS reaction.