Kinetic study of hydrogenation reaction in a slurry-phase hydrocracking of vacuum residue oil

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A kinetic model was developed for catalytic hydrogenation reactions of vacuum residue in a CSTR with a commercial slurry-phase catalyst (Mo-octoate). The kinetic model included hydrodesulfurization, hydrodenitrogenation, hydrodemetallization, hydrode —asphaltenization, hydrodemicrocarbonization. The experiments were conducted at the severe operating conditions with temperature range of 410 – 450 °C, liquid hourly space velocity (LHSV) of 0.25 – 1 hr-1, hydrogen pressure of 160 bar, gas to oil ratio (GOR) of 1500 Nm3/m3, and concentration of Molybdenum of 500 ppm. The developed kinetic model accounted for mass transfer and vapor-liquid equilibrium. The developed kinetic model was well in agreement with experimental data.