

NO<sub>x</sub> storage over Pd/CeO<sub>2</sub> and Pd/SSZ-13 for cold start applications

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Automobile after-treatment systems are being developed to meet the regulations. Therefore, state-of-the-art exhaust catalysts work effectively to convert most pollutants. However, pollutants released during the cold-start period, when the catalytic converts have not reached the proper operating temperature, are not effectively converted. Passive NO<sub>x</sub> adsorbers (PNA) are attracting attention as a potential solution. The PNA can store the NO<sub>x</sub> emitted during the cold-start period and then release the NO<sub>x</sub> once the catalysts are sufficiently heated to convert NO<sub>x</sub>. Metal oxides and zeolites combined with precious metals have been widely studied as the PNA materials. It is advantageous to compare metal oxide based PNAs and zeolite based PNAs because they operate in a similar manner but with different NO<sub>x</sub> adsorption and desorption mechanisms. In this work, Pd/CeO<sub>2</sub> and Pd/SSZ-13 are compared by various methods. Adsorption sites and mechanisms were identified using different NO/NO<sub>2</sub> ratios for adsorption tests. Effect of CO treatment and sulfur poisoning was also investigated on both samples.