## Layered Double Hydroxide Derived Intermetallic $Ni_3GaC_x$ Catalysts for Dry Reforming of Methane

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NiMgGa-layered double hydroxide (NMG-LDH) was synthesized as an efficient catalyst precursor for dry reforming of methane which shows high stability at low temperature (~600°C). NMG-LDH was transformed to Ni<sub>3</sub>Ga intermetallic structure after the reduction pre-treatment, which exhibited high CH<sub>4</sub> (~48%) and CO<sub>2</sub> (~52%) conversion as well as high stability compared to other reported monometallic Ni-based catalysts under similar reaction conditions. The stability of Ni<sub>3</sub>Ga intermetallic catalysts was superior to that of monometallic catalysts (Ni/MgO or Ga/MgO) due to high coke resistance property. The formation of unique intermetallic carbide (Ni<sub>3</sub>GaC<sub>x</sub>) during the reaction was responsible for the outstanding stability compare to the conventional Ni monometallic phase. The formation of Ni<sub>3</sub>GaC<sub>x</sub> intermetallic carbide structure was confirmed by XRD, HR-TEM, and XAS analysis. *In situ* XRD analysis was investigated to elucidate formation mechanism of Ni<sub>3</sub>GaC<sub>x</sub> structure and reaction mechanism on the Ni<sub>3</sub>GaC<sub>x</sub> during the reaction. Carburization of Ni<sub>3</sub>Ga by CH<sub>4</sub> and oxidation of Ni<sub>3</sub>GaC<sub>x</sub> by CO<sub>2</sub> make a redox cycle, resulting in high activity and stability.