## Investigation into the start-up solvent for slurry phase Fischer-Tropsch synthesis

<u>이경우</u><sup>1,2</sup>, 임근배<sup>2</sup>, 이현송<sup>2</sup>, Deviana Deviana<sup>2</sup>, 윤민혜<sup>2</sup>, 이관영<sup>1</sup>, 천동현<sup>2,†</sup>

<sup>1</sup>고려대학교; <sup>2</sup>한국에너지기술연구원

(cdhsl@kier.re.kr<sup>†</sup>)

Fischer–Tropsch synthesis (FTS) is considered an attractive way to produce valuable hydrocarbon products from low-value carbon–containing resources such as coal, natural gas, waste biomass/plastics, and CO<sub>2</sub> via syngas (CO + H<sub>2</sub>). A slurry phase reactor can be efficiently used for FTS because this reactor can easily remove the heat released from catalysts to heat exchanger via a liquid medium, which allows near isothermal operation of FTS. A liquid medium is essential for carrying out the FTS in a slurry phase reactor. In particular, an initial liquid medium should be included in the slurry phase reactor. Therefore, the start-up solvent can play an important role in determining catalyst performance in the slurry phase FTS. In this study, we performed the FTS in a continuously stirred tank reactor (CSTR) using various start-up solvents such as mineral oil, icosane, and squalene. We focused on the effects of start-up solvents on the activity, stability, and selectivity of precipitated iron-based catalysts. The activity and stability were evaluated in terms of CO conversion as a function of reaction time, and the selectivity was evaluated in terms of hydrocarbon distribution.