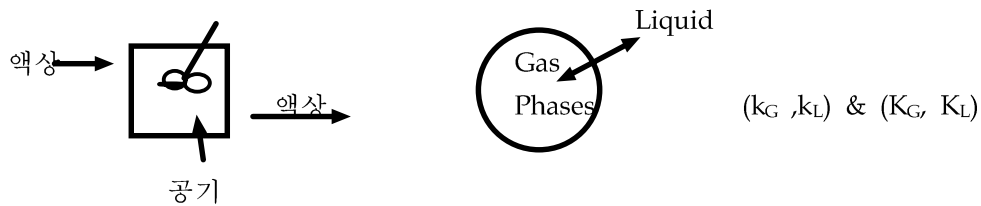


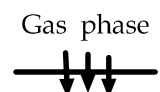
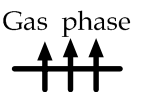
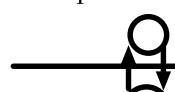
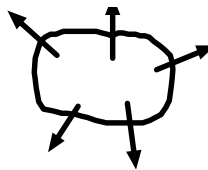
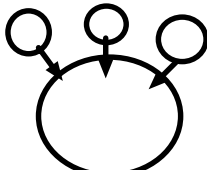
Chap. 31. Mass-Transfer Equipment

- Design of continuous-contacting mass transfer equipment
 - design for new equipment or improvement in performance
 - final design equations in terms of overall driving force and overall transfer coefficient
 - total contact area within mass exchanger changing the compositions using interphase mass-transfer principles
- Convective mass transfer (theory, experiment)
- Driving force (농도차) → flux → contacting area
- Ex: Interphase mass transfer (bubble)



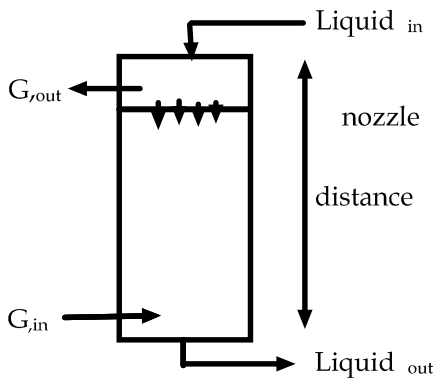
1. Types of mass-transfer equipment

- transfer of a solute from the gas phase into the liquid phase: absorption, dehumidification, distillation :
- transfer of a solute from the liquid phase into the gas phase: desorption, humidification
- transfer of a solute from one liquid phase into another immiscible liquid phase: liquid-liquid extraction, ex.) Purification of toluene (toluene+benzoic Acid) using water
- transfer of a solute from a solid into a fluid phase: drying, leaching
- transfer of a solute from a fluid onto the surface of a solid: adsorption, ion exchange

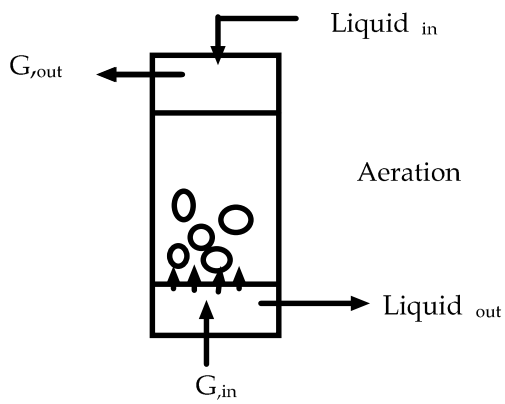
I	II	III	IV	V
Gas phase  Liquid phase	Gas phase  Liquid phase	Liquid I  Liquid II	 고형물로부터 액상물질의 기화	 고형물 표면위로 물질이 달라붙음

· Intimate contact of the two phases : 중력이용

i) Spray tower

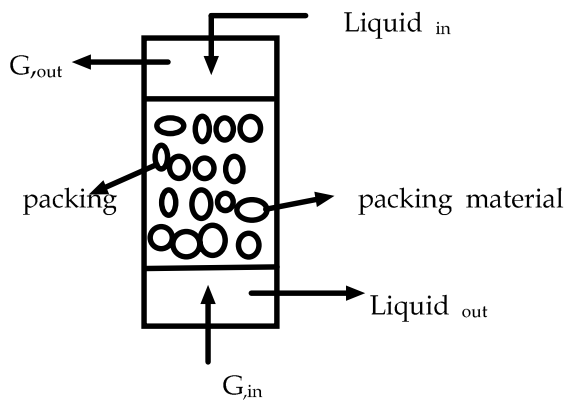


ii) Bubble tower



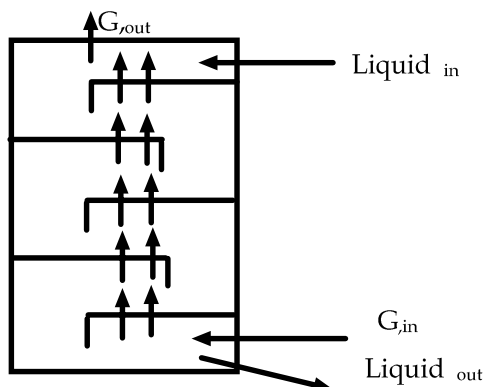
- 작을 경우
- solute a highly soluble gases
NH₃, SO₂

iii) Packed tower



large interphase contact area

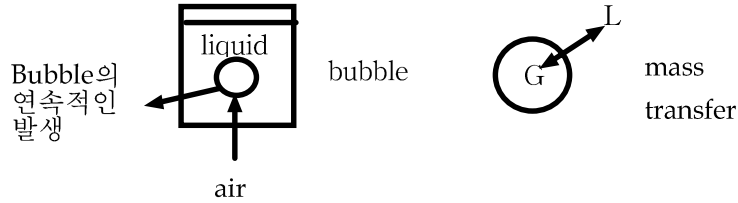
iv) Bubble plate tower



stagewise
(stage 별 평형관계 도출)

2. Batch mass-transfer tank or ponds

- Waste-water treatment undesirable gas → stripping or desorbing
- 산소공급 (Oxygen) → aeration



- interphase mass transfer (concentration driving force)
- slightly soluble gas into liquid

$$\text{mass transfer rate : } N_A = K_L (C_A^* - C_{A,L})$$

$$C_{A,L} \text{ 측정, } P_{A,G} \text{ 측정} \rightarrow C_A^* \text{ 환산}$$

$$K_L = K_L (k_L, k_G \text{ \& } m) \quad (m : \text{Henry 상수})$$

$$W_A = K_L A (C_A^* - C_{A,L}) = d(\text{mole of } A) / dt = \frac{V d C_{A,L}}{dt}$$

$$\frac{d C_{A,L}}{dt} = \frac{K_L A}{V} (C_A^* - C_{A,L}) \rightarrow \frac{d C_{A,L}}{(C_A^* - C_{A,L})} = \frac{K_L A}{V} dt$$

$$t = 0 \rightarrow t = t$$

$$C_{A,L} = 0 \rightarrow C_{A,L,t} \text{ 을 범위로 적분}$$

$$\ln \left[\frac{C_A^* - C_{A,L}}{C_A^* - C_{A,L,t}} \right] = K_L \frac{A}{V} t$$

$$K_L \frac{A}{V} = \frac{\theta_g Q_g^{1+n} h^{0.78}}{V}$$

- θ : const , θ_g : gas flow rate, h : depth, V : liquid의 부피