

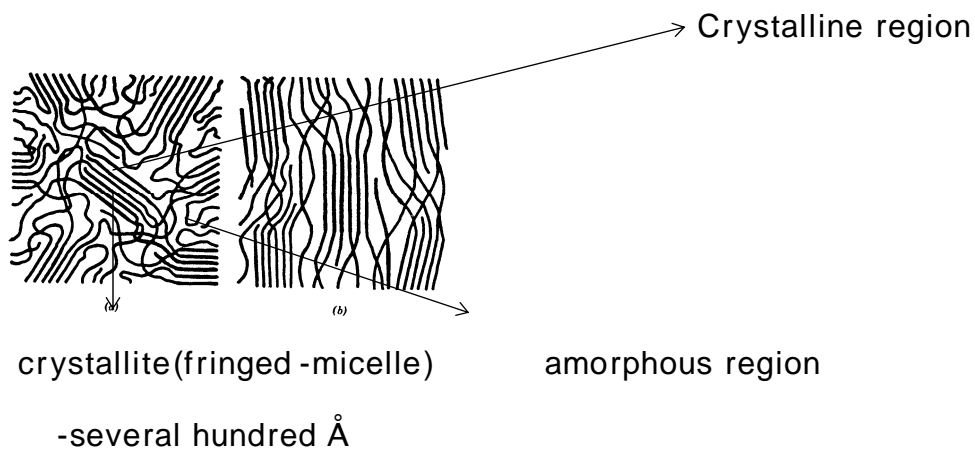
## Chapter 5. Crystallinity

### 5.1 Requirements for crystallinity

- Regular chain structure. ex) iso -PP, iso -PS
- Hydrogen bonding
- Strong dipole interaction. ex) nylon -6

### 5.2 The Fringed -Micelle Model

- solid state polymer
  - i) completely amorphous
  - ii) partially crystalline
  - iii) almost completely crystalline.

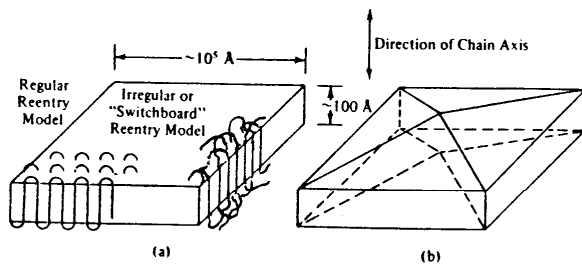


If they are stretched → increase crystallinity.

- crystallites tie the individual chains together
- the crystallites will generally melt before the polymer degrades

### 5.3 Folded -chain Crystallites

- the growth of single crystals from dilute solution



- i) regular reentry model
- ii) adjacent reentry model
- iii) irregular or “switchboard” reentry model

- growth rate 가 가 more perfect crystal 가 .
- Chain folding why? 1000 Å 가 100 Å folding .
- Solution state : lamellae (plate like polymer crystal)
- Melt state : a model combining the folded -chain lamellae with the interlamellar amorphous material.

#### 5.4 The effects of crystallinity on polymer properties

- LDPE (low density polyethylene)
  - long branched branches : made by high -pressure process. ( 25,000 - 50,000 psi)
  - $\delta=0.915\text{g/cm}^3$  42 -53% crystallinity  $T_m= 110 -120$  °C
- LLDPE (linear low density polyethylene)
  - short, straight branch : low -pressure (100 psi)
  - $T_m= 120 -130$  °C 54 -63% crystallinity

- HDPE (high density PE)
  - no branches, linear : made by low pressure process  
 $\delta=0.97 \text{ g/cm}^3$       64 -80% crystallinity
  - more tightly packed in the crystalline than in the amorphous areas.  
 e.g.) iso -PP: crystalline, hard and rigid plastics.
- crystalline polymers have two -phase systems with a crystalline phase dispersed in an amorphous matrix - mostly opaque.
- In general, transparent polymers are completely amorphous.

## 5.5 Determination of Crystallinity

### 1. density measurement

$$\phi_c = \frac{\rho - \rho_a}{\rho_c - \rho_a}$$

(volume fraction of material in the crystalline state)

$\rho$  : density of the sample

$\rho_a$ : density of amorphous polymer

$\rho_c$ : density of crystalline polymer – X -ray unit cell dimension.

$$w_c = \frac{\rho_c(\rho - \rho_a)}{\rho(\rho_c - \rho_a)}$$

### 2. Specific heat (cal/g·°c)

$$\chi = \frac{c_a - c}{c_a - c_c}$$

### 3. melting enthalpy (enthalpy of fusion)

$$\chi = \frac{\Delta H}{\Delta H_f}$$

#### 4. IR

$$\chi = \frac{\epsilon_{\lambda}}{\epsilon_{\lambda_c}}$$

e.g.) P.E : 7.67 $\mu$ m      amorphous.

#### 5. NMR

Broad band : crystalline regions

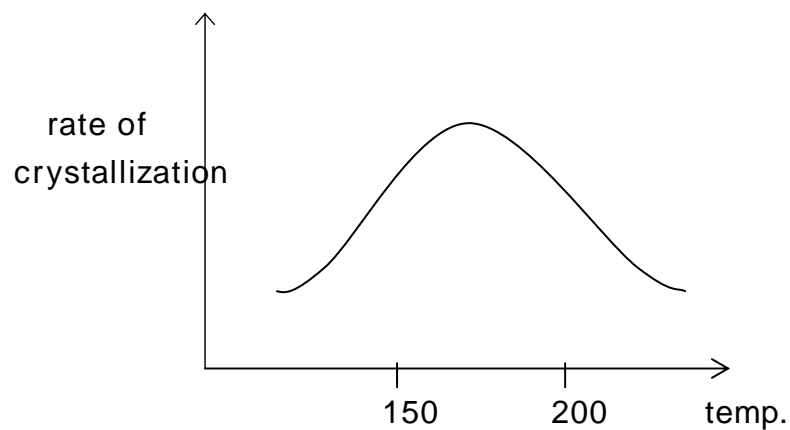
Sharp band : amorphous regions

#### 6. X-ray

$$\chi = \frac{I_c}{I_c + I_a}$$

$I_c$  = integrating the intensities of crystalline reflections.

. Rate of crystallization



- rate of crystallization is maximum between  $T_g$  and  $T_m$  of the polymer.

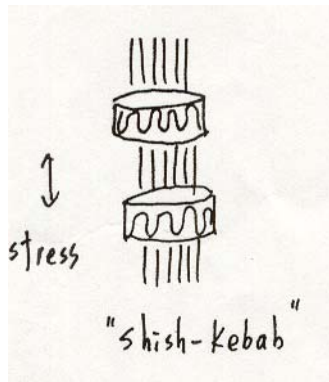
. Secondary crystallization

quenching of specimen at melt temperature and then store at temp. higher than  $T_g$

→ the disordered regions will be mobile enough to rearrange into lower energy, more ordered structure.

#### 5.6 Extended-chain crystals

. fibrillar structure – polymers crystallized from a melt while subjected to extensional flow (extended – chain crystals)



#### 5.8 Spherulites ( )

. polycrystal                      single crystal                      . – spherulites are aggregates of lamellar crystallites.

)

- i) grow radially from a point of nucleation.

- ii) nuclei가                      spherulite 가                      . Shock cooling → smaller spherulites.

- iii)                      0.01mm(diameter) - "Maltese Cross"

- iv) semicrystalline polymer .

