

- has better chemical resistance than PS,
 - higher heat resistance
 - stiffer than PS
 - not as clear as PS
-
- Acrylonitrile - butadiene - styrene (ABS) copolymer
 - tough plastic
 - good mechanical property
 - chemical resistance 가
 - good electrical property ()

Thermosetting and Engineering Resins

(1) Thermosetting Resins ()

- 가 가 , 가 flow가 (degradation)가 .

- a network of long chain molecules that are crosslinked which gives the polymer a three -dimensional, infusible structure.

(ex) Epoxy resin, phenol resin, polyurethane, unsaturated polyester resin

(2) Engineering Thermoplastics -(가)

- high thermal property(stability),
- good chemical and weather resistance ()
- good electrical property ()

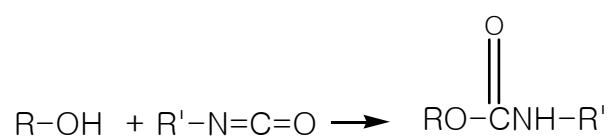
(ex) Nylon, PET, PBT, Polycarbonate, polyacetal, PMMA

(1) Themosets

(a) Polyurethanes (PU)

- reaction between alcohol (or diol) and isocyanates
- produced by condensation reaction.

()

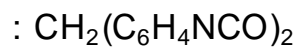


urethane linkage

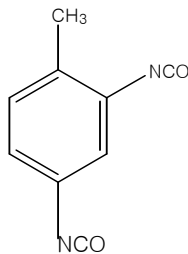
- polyurethane isocyanate

MDI TDI가

. MDI (4,4 -methylene bis(phenyl isocyanate))



. TDI (2,4 -and 2,6 -toluene diisocyanate)



()

rigid foam ()

flexible foam ()

adhesive ()

surface coating

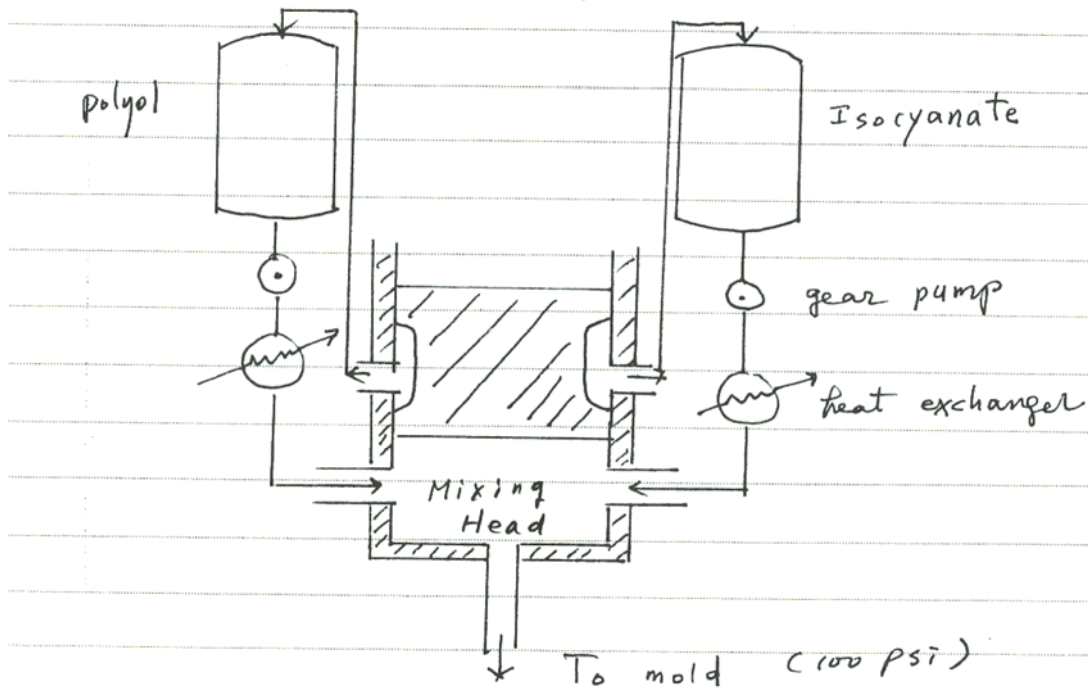
elastomer

Reaction Injection Molding(RIM)

-
- a process developed recently to mold large polyurethane directly from the starting chemicals.
- the mixed stream begins to react as it flows into the mold, where the polymerization reaction is completed.

() highly reactive polyol isocyanate.

- nozzle mixing head .



. The polyurethane maybe formulated – flexible rigid PU,
solid foamed PU.

. : energy – absorbing front and rear panels for automobiles.

Insulating materials(,)
: polyurethane foam
LNG tank (rigid PU foam)

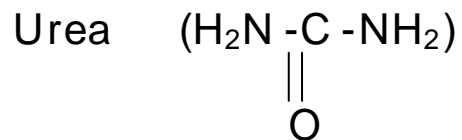
Reinforced RIM (RRIM)

- reinforced materials, in which short reinforcing fibers (glass fiber :) are incorporated or both of the

reactants.

-

(1cm).

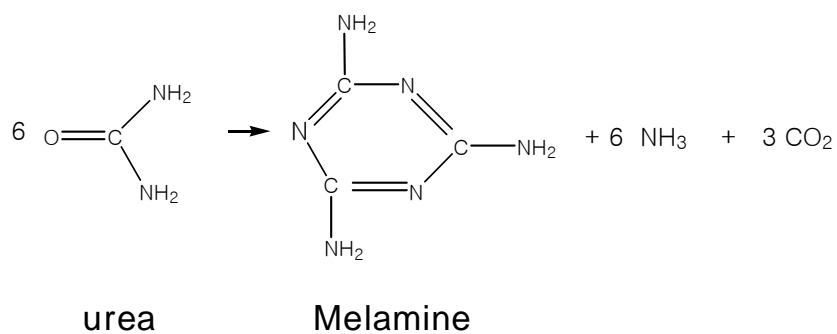


- produced by the reaction between ammonia and carbon dioxide.

Urea-formaldehyde resins : the condensation reaction between urea and formaldehyde.

Melamine – formaldehyde resins :

- melamine is a heterocyclic aromatic condensation product of urea



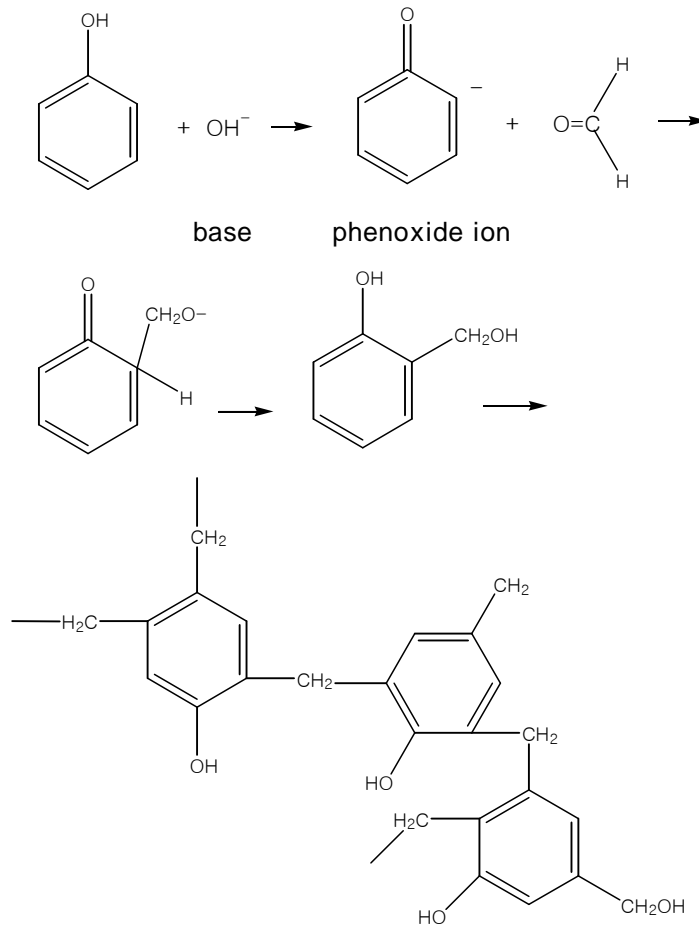
()

Phenolic Resins

- phenol formaldehyde phenol ,

- condensation .

()



(a) hardness and rigidity

(b) acid and water hydrolysis resistance

(c)

(d) 150 가

(e) , 가

(f)

Epoxy Resins

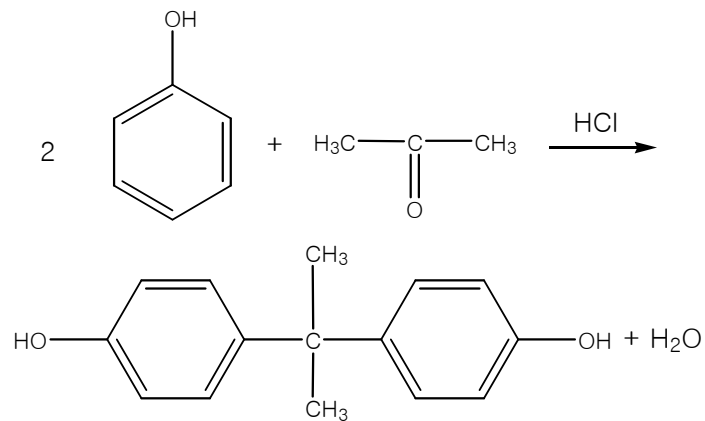
(1) Phenol acetone bisphenol -A

(2) Bisphenol -A epichlorohydrin
diglycidyl ether of bisphenol -A(DGEBA)

(3) DGEBA (epoxy resin) amine
crosslinked epoxy .

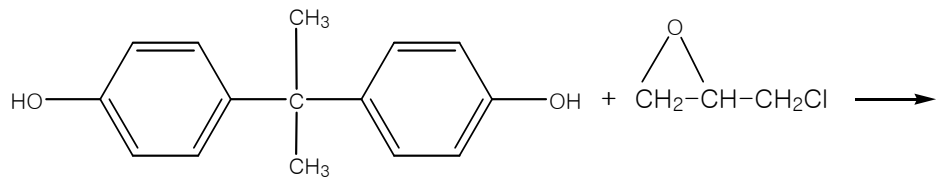
- 가

(1)



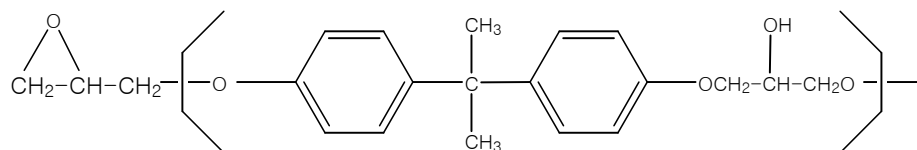
Bisphenol -A

(2)

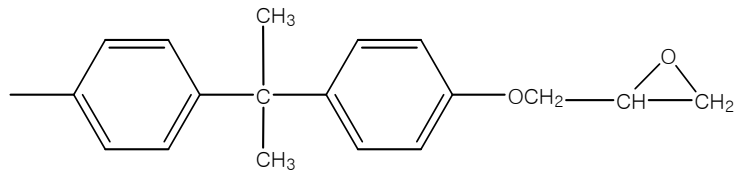


(BPA)

(epichlorohydrin)



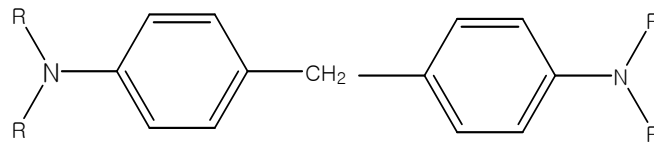
epoxy functional group



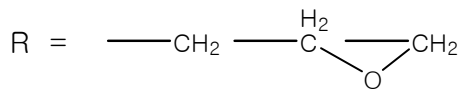
Diglycidyl ether of bisphenol A (DGEBA)

- DGEBA epoxy function group 2 Bisphenol -A
type epoxy

- epoxy functional group 4 Novolac
type epoxy



..

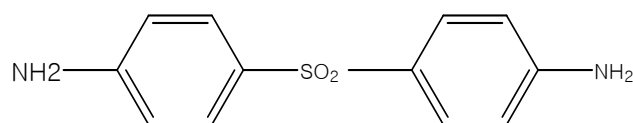


Tetra glycidyl diamino diphenyl

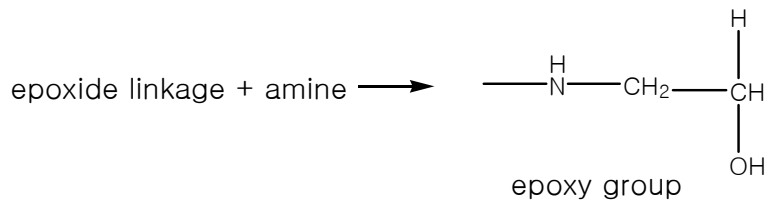
methane(TGDDM)

- DGEBA TGDDM linear
polymer . epoxy
polymer hydroxyl group epoxide
linkage amine .

- (curing agent) .

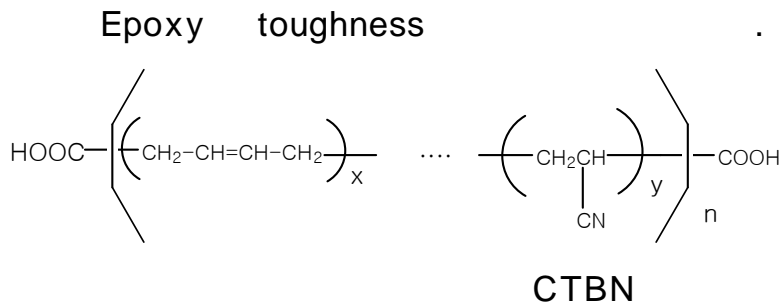


Diaminodiphenyl sulfone (DDS)



- (a) metal .
- (b)
- (c) 250~260 가 .
- (d) Good dimensional stability (),
 , low shrinkage property

Epoxy brittle carboxyl -terminated
butadiene acrylonitrile(CTBN) liquid rubber 가



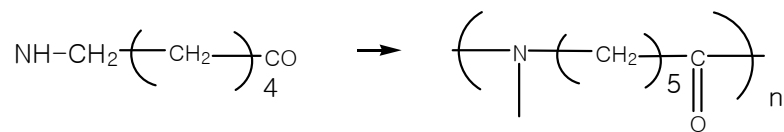
(reference) Kim, Park and burns,
Journal of Applied Polymer Science, Vol. 50. page 1951~1957
(1993)

Engineering Polymers

(1) Nylon

() Nylon -6 (polycaprolactam)

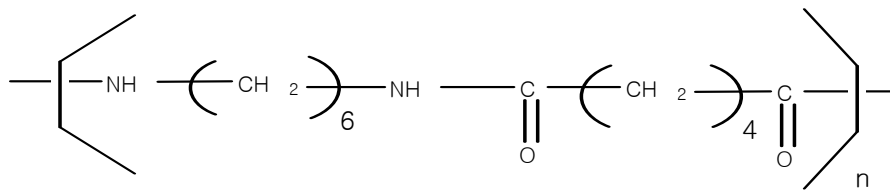
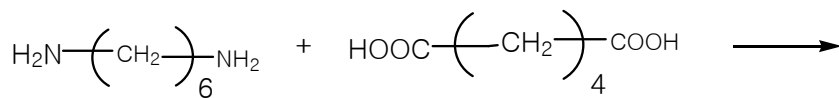
- produced by polymerization of caprolactam (condensation reaction).



() Nylon -6,6 (polyhexamethylene adipamide)

- produced by polycondensation of hexamethylene diamine and adipic acid.

()



()

() High tensile and impact strength

() toughness, good abrasion and wear resistance()

()

(a) Blow and injection molding

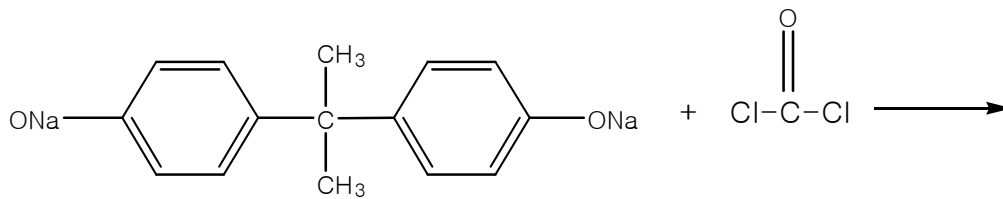
(b) Bearings, molded gears, molded automobile parts(), and

machinery parts

(c) Tubing, wire and cable insulation()

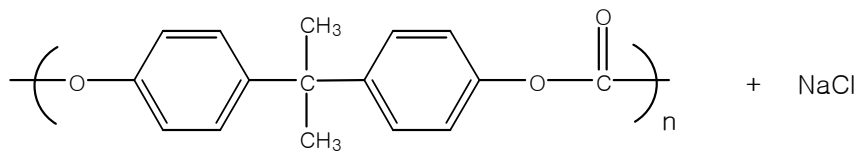
(d) thermoplastic engineering resin.

(2) Polycarbonate(PC)



sodium salt of Bisphenol -A

phosgene



Bisphenol -A polycarbonate

()

(a) (transparent), break and heat resistant

(b) very tough, self extinguishing ()

(c) good electrical property()

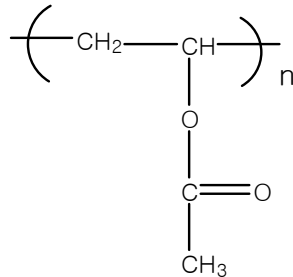
()

(a) Injection, blow molding, extrusion

(b) Lab safety shields, street lighting globes

(c) Safety helmets, sun glasses, school windows

(3) Polyvinyl acetate :



- produced from ethylene and acetic acid
- free radical
- the polymer is highly branched, amorphous and atactic

()

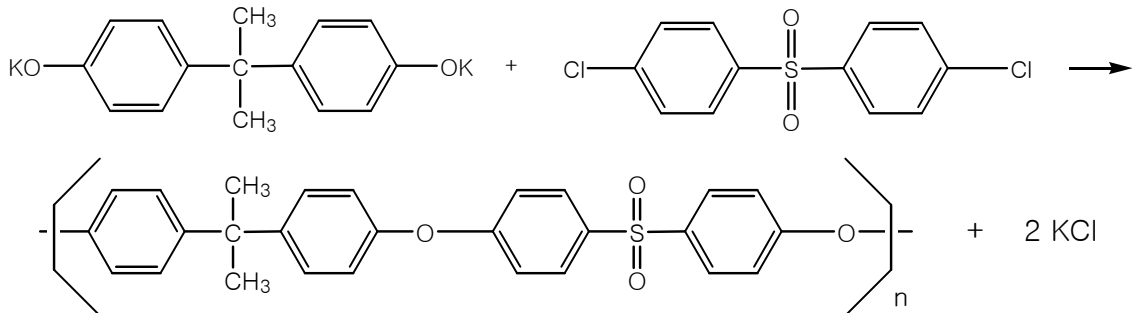
(a) colorless, odorless and nontoxic

()

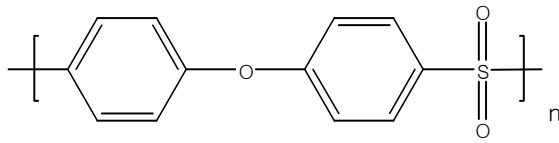
(a) polyvinyl alcohol

(4) Aromatic polyethersulfone (PES)

- prepared by sodium or potassium salt of bisphenol -A and 4,4 -dichlorodiphenyl sulfone



(: Udel : Union carbide)

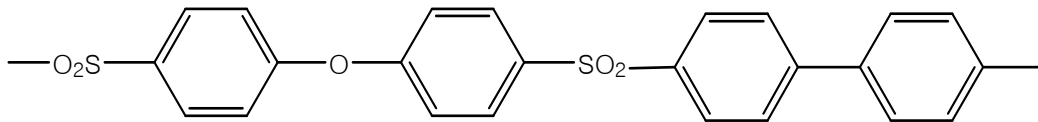


: PES 200P (by. ICI)

PES

.Fig. 13-10 shows the maximum use temperature for several engineering plastics

(5) Polyaryl sulfone



: Radel (Union Carbide)

()

(a) Resistant to alkalis, acid and inorganic salts, but not resistant to highly polar chlorinated hydrocarbons

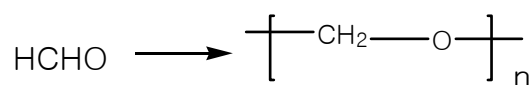
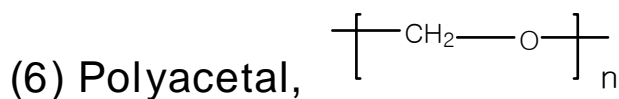
(b) Similar to those of PC

(c) Can be used at higher temp.

()

(a) extruded into thin films and foils

(b) injection molded into various objects which need high temp stability.



(form aldehyde) polyoxymethylene (polyacetal)

()

(a) highly crystalline with good dimensional stability

(b) High impact resistance

(c) great strength

(d) resistant to chemical and hydrolysis

()

(a) door handles molding

(b) gears, electronic equipment

(c) appliances, (기) parts

(7) Thermoplastic. polyesters

(a) PET

(b) PBT : 1,4 -butandiol + Terephthalic acid

()

electronics parts

sporting goods, automobile bumper

(ex) PBT + PC -> "Xenog Bumper"

Blends

Chapter 14. Synthetic Fibers

- Fiber

(1) high melting point (100)

(ex) PET : $T_m=253$

Nylon -6 : $T_m=230$

(2) linear and symmetrical structure

(3) high molecular weigh

(ex) PET, $M_w = 30,000$ (g/mol)

Nylon -6, $M_w=35,000$ (g/mol)

(synthetic fiber)

Polyester(PET) 49%

Nylon 33%

Acrylic 9%

Olefin 9%

100%

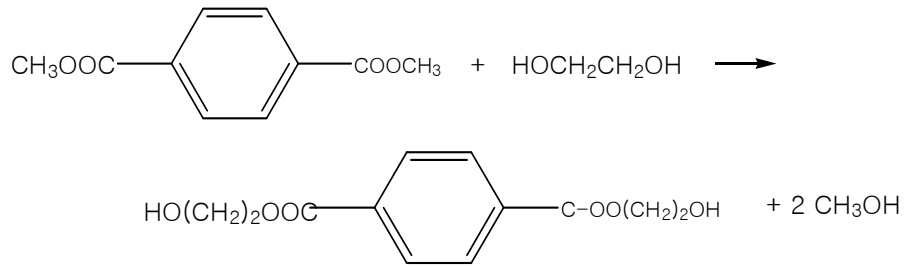
Polyester Fibers

-Pet(Polyethylene terephthalate) :

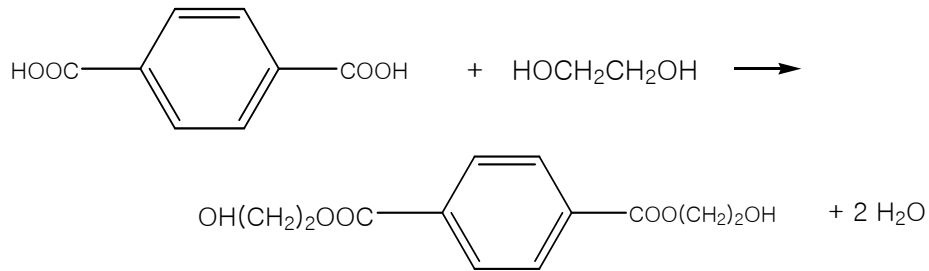
()

() dimethyl terephthalate (DMT) + ethylene glycol (EG)

-> PET



() terephthalic acid + ethylene glycol (EG) > PET



- PET cotton blending 가

- PET staple fiber/cotton = 50: 50 or 65:35

Polyamides

- Nylon :

condensation polymers formed by the reaction between :

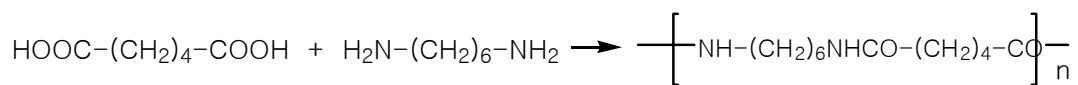
(a) dicarboxylic acid a diamine : Nylon -6,6, 610

(b) a ring opening of lactams : Nylon6,12

(c) polymerization of amino acids : Nylon -7,11

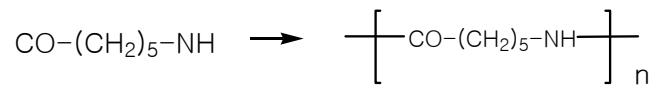
.Nylon -6,6(polyhexamethylene adipamide)

adipic acid + hexamethylene diamine -> Nylon -6,6



Nylon6,6

. Nylon -6 (polycaprolactam)



()

(a) linear polymer and highly crystalline

(b) Nylon -6 $T_m=230$

Nylon -11 $T_m=190$

(c) Nylon -4 moisture absorption cotton

(d) Nylon .

()

(a) tire cord

(b) carpet . ropes, parachutes, selt belts, conveyor

belts

Acrylic Fibers

- the acrylics are the third major class of synthetic fibers.

- polyester

- polyacrylonitrile

- temperature sensitive melt spinning solution spinning .

- solvent dimethylformaide(DMF) .

modacrylic fiber

- the fiber -forming substance is any longchain synthetic polymer

composed of less than 85% but at least 35% by weight of acrylonitrile units, $-(\text{CH}_2\text{CHCN})-$ chemicals vinyl chloride, are incorporated as modifiers.

()

(a) tradename Orlon, Acrilan

(b) highly resistant

(c)

Polypropylene Fibers

isotactic PP(I-PP) : $T_m = 171$

= 60~85%

()

(a) high abrasion resistance

(b) strength, low static build up

(c) resistance to chemicals.

(d) low softening point

(e) low resilience and dyeing problems

(f) carpet

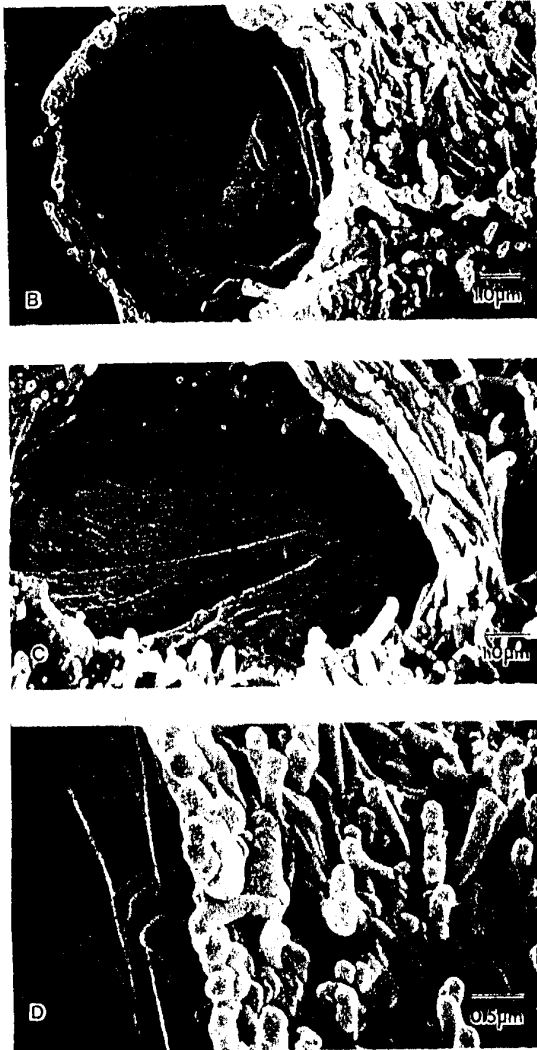


Fig. 5.96 A glass fiber reinforced LCP composite is shown to have interesting morphology. A polished thin section is shown in polarized light (A) (color section) to exhibit a fine domain texture with some orientation of the polymer on the glass surfaces. SEI fracture views (B-D) show the tenacious adhesion of the LCP to the fibers. Fibrillar structures are oriented parallel to the fiber surface and submicrometer sized domains are observed (D).

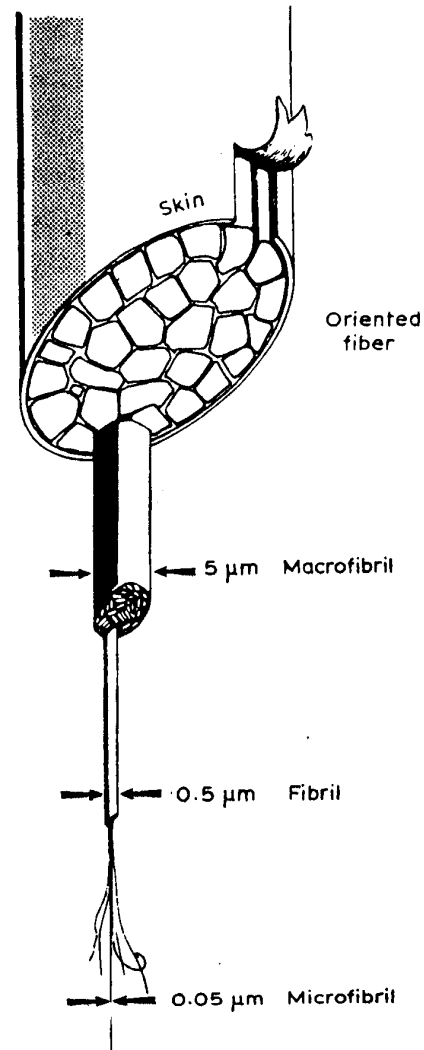


Fig. 5.97 The structures observed in uniaxially oriented LCP fibers, ribbons and films can be summarized by this structural model. The model defines the nature of the fibrillar textures into three categories based upon size: macrofibrils, fibrils and microfibrils. In each case the sizes of structures have been determined from complementary microscopy techniques. (From Sawyer and Jaffe [353]; reproduced with permission.)

(ref.) *J. Mater. Sci.*,
21, 1897 (1986).

(reference) "polymer Microscopy", by Sawyer and Grubb,
Chapman and Hall, 1987, page 254

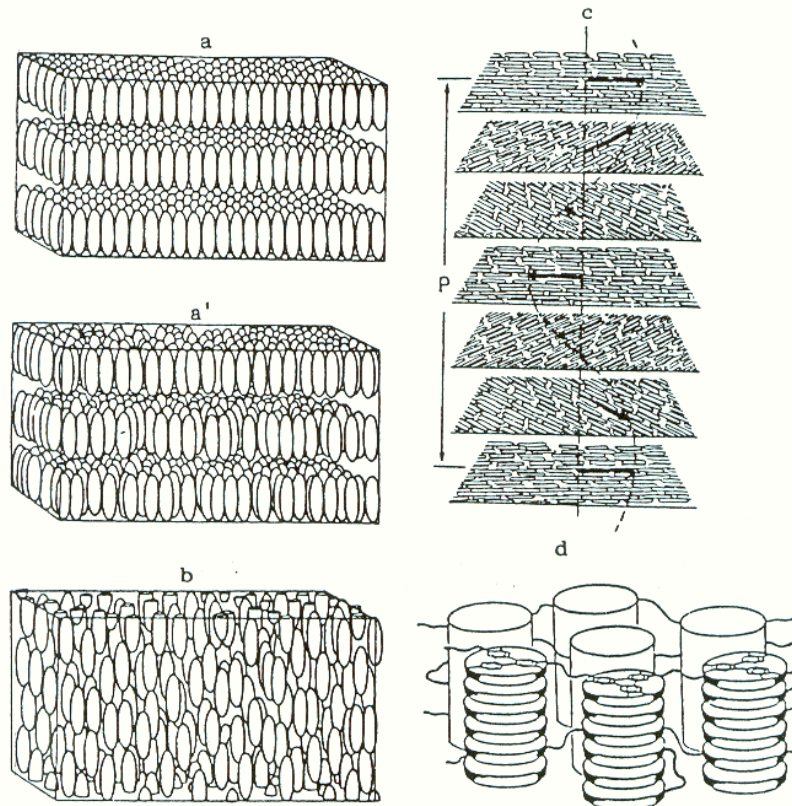


Fig. 2.17. Schematic representation of the different types of mesophases. Smectic with ordered (a) and unordered (a') arrangement of the molecules in layers; b) nematic; c) cholesteric; and d) discotic. (From Platé and Shibaev (1987); Courtesy Plenum Press).

order (from kheir, hand, and from kholè and stereos: bile and firm); sometimes a fourth category is distinguished: the *discotic* one (piles of dislike molecules). Fig. 2.17 gives a schematic representation.

If monomeric liquid crystals are polymerized, the polymers also show in most cases the liquid-crystalline effects. In addition, their viscosity during flow is unexpectedly low in the anisotropic phase state.

There is no consensus yet as far as the name of these materials is concerned. Some investigators use the name *Polymer(ic) Liquid Crystals (PLC's)*, others call them *Liquid Crystalline Polymers (LCP's)* or *Mesogenic Macromolecules*.

PLC's which melt without decomposition are called *thermotropic*; those who decompose on heating before melting but can be dissolved in liquids, are called *lyotropic*.

(ref) "Properties of Polymers", by van Krevelen, Elsevier, 2nd ed(1990), page, 35

- (a) ordered smectic

(b) unordered smectic

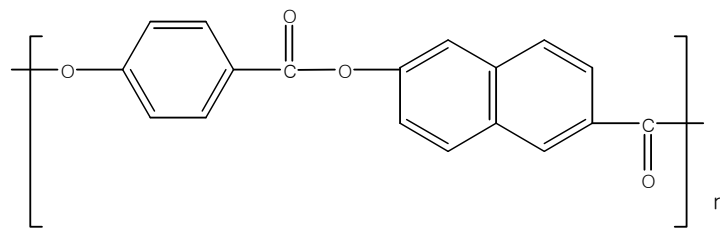
(c) nematic

(d) discotic

Liquid Crystalline Polymer (LCP)

) thermotropic LCP - , 가
crystal -to -nematic transition

ex) Hoechst -Celanese Vectra A900



p -HBA

HNA

(hydroxy benzoic acid)

6hydroxy -2 -naphthoicacid

73mol%

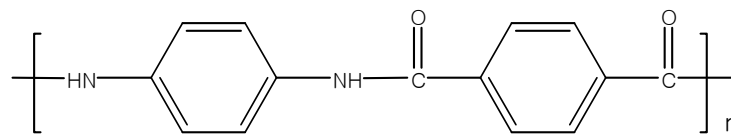
27mol%

Du pont, Eastman Kodak

Amoco

) lyotropic LCP

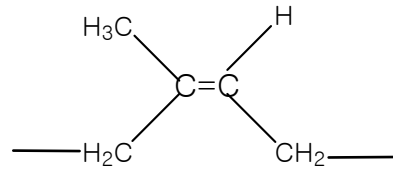
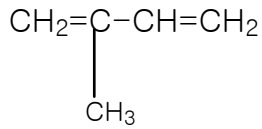
ex) DuPont Kevlar



high modulus fiber (106 kg/cm²)

Synthetic Rubber

.Natural Rubber - composed mainly of high molecular weight
hydrocarbon made up of isoprene,

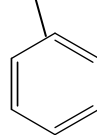
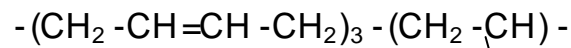


- . The isoprene units are linked in a cis -1,4 configuration
 - high resilience, strength
- . Elastomers (), synthetic rubbers, are polymers with physical and mechanical properties similar to those of natural rubber.
- . Elastomers are unsaturated and can be vulcanized.

Synthetic rubber

(a) Butadiene rubber, $\text{CH}_2=\text{CHCH}=\text{CH}_2$

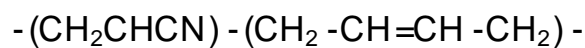
(b) SBR(styrene -butadiene -rubber)



75mol%

25mol%

(c) ABR or NBR (acrylonitrile -butadiene rubber)



(d) Polyisoprene

(e) Transpolybutadiene Rubber (TPR)

(f) Ethylene -Propylene Rubber (EPR)

(g) Polychloroprene Rubber (Neoprene)

(a) Polybutadiene Rubber(PBD)

- the most important synthetic rubber
- monomer (copolymerization) 가
- free radical initiator 1,4 -PBD , styrene

SBR

- acrylonitrile ABR(or NBR)
- natural rubber ABR blending 가

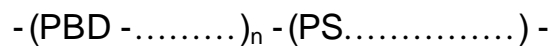
()

- 95% tire industry
- high abrasion resistance()
- high resilience
- cis trans Tg가

(ex)

- high cis(98%) 1,4 -PBD (Tg=-108)
- medium cis (52%) 1,4 -PBD (Tg=-80~ -96)
- trans PBD (Tg=-14)

Thermoplastic Elastomer(가)



Poly(styrene -b -butadiene) block copolymer

- the poor mechanical strngth of cisPBD is improved by incorporating styrene block copolymer.

(b) Styrene-butadiene rubber (SBR)

- tire industry
- good mechanical and physical properties
- favorable cost
- made by copolymerization of ca 75% butadiene and 25% styrene
- Styrene-butadiene-styrene (SBS) triblock copolymer

TPE

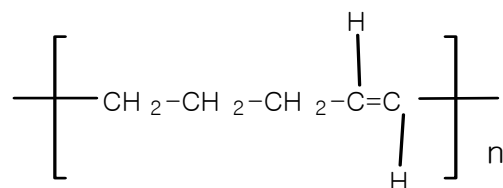
(c) ABR or NBR (acrylonitrile-butadiene styrene)

- produced by copolymerizing varying proportions of butadiene and acrylonitrile in an aqueous emulsion.
- NBR is resistant to hydrocarbon liquids
- fuel hoses, fuel cell liners, gaskets

(d) Polyisoprene

- produced commercially by the polymerization of isoprene using either lithium and lithium compounds as catalysts
- butadiene blending

(e) Trans polybutadiene (TPB) rubber;



- high abrasion resistance ()

- high mechanical properties - excellent processibility

(f) Ethylene-propylene rubber (EPR)

- ethylene-propylene copolymer crosslinking
double bond가 , 1,4-hexadiene 3
monomer
- elastomer ethylene-propylene terpolymer (EPT) or
ethylene propylene diene monomer rubber (EPDM)
- good abrasion property
- good heat resistance
- slow cure rate
- SBR SBR/EPR blending
()
tire, cable coatings, appliance parts

Table 21.1 Radial Passenger-Tire Tread Formulation

Ingredient	phr ^a	Function
SBR 1712^b		
Polymer	45	Rubber polymer
Aromatic oil	37.5	Extending oil
<i>cis</i> -Polybutadiene 1252	55	Rubber polymer
Carbon black N-234	70	Reinforcing agent
Oil-soluble sulfonic acid	1	Processing aid
Sulfur	1.75	Vulcanizing agent
Stearic acid	2	Promotor
Zinc oxide	3	Promotor
<i>N</i> -Cyclohexyl-2-benzothiazole sulfenamid	1	Accelerator
Polymerized 1,2-dihydrotrimethylquinoline	2	Antioxidant
<i>N</i> -(1,3-dimethylbutyl)- <i>N'</i> -Phenyl- <i>p</i> -phenylene diamine	1	Antiozonant
Blended petroleum wax	3	Crack inhibitor and antiozonant
Total	222.25	

^a Parts per hundred parts rubber, by weight.

^b Oil-extended master batch. Cold emulsion polymer \approx 75% butadiene, 25% styrene extended with highly aromatic oil.

Source. Stephens.⁷