

7. Reaction Mechanisms, Pathways, Bioreactions and Bioreactors

- Active Intermediates and Free Radicals
 - Pseudo-steady-state-hypothesis
 - Reaction pathway
- Enzymes
 - Michealis-Menten enzyme kinetics
 - Enzyme inhibition
- **Bioreactors**
- **Pharmacokinetics**
- **Polymerization**

2. Enzymes XVIII

○ Types of Enzyme Inhibition 10

- Substrate inhibition

- most substrate acts as an inhibitor



- **incompetitive reaction, I → S**

$$-r_S = \frac{V_{\max} S}{K_M + S + \frac{S^2}{K_I}}$$

- **at low conc. of substrate**

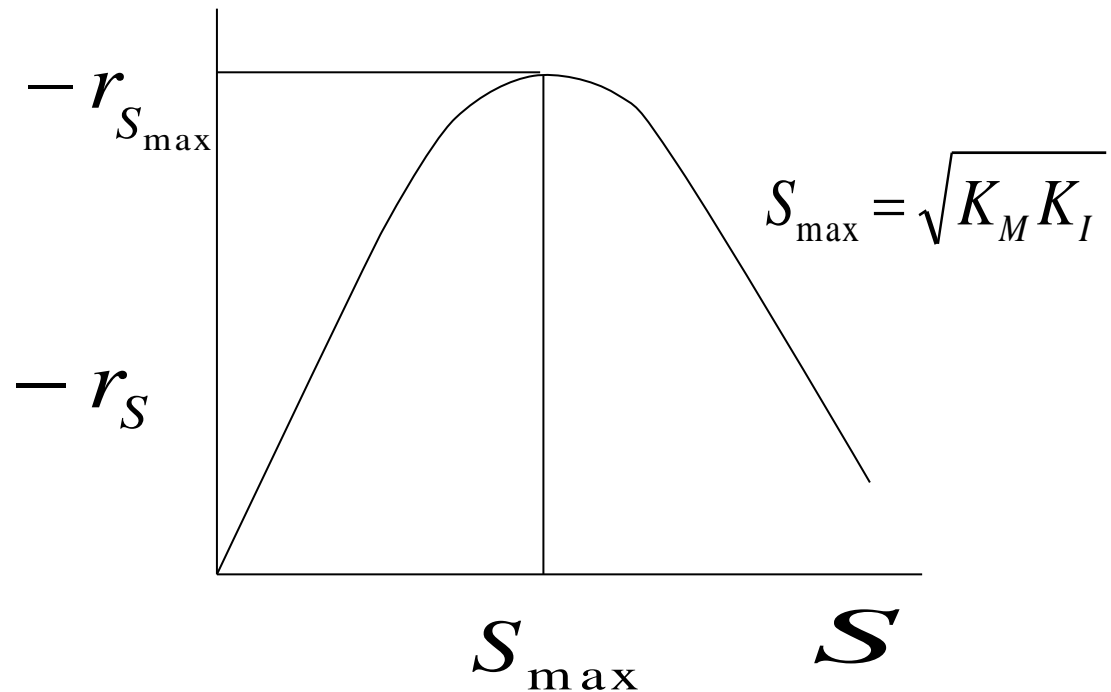
$$K_M \gg \left(S + \frac{S^2}{K_I} \right) \quad -r_S \sim \frac{V_{\max} S}{K_M}$$

2. Enzymes XIX

- Types of Enzyme Inhibition 11
 - Substrate inhibition 2
 - at high conc. of substrate

$$K_M + S \ll \left(\frac{S^2}{K_I}\right)$$

$$-r_S = \frac{V_{\max} K_I}{S}$$



3. Bioreactors I

○ Bioreactor

- A reactor that sustains and supports life for cells and tissue cultures
- Cell metabolism
 - transformation of chemical energy
 - construction, breakdown, digestion of cellular components
- Microbial growth
 - enzyme rxns are involved during growth
- Monod equation  similar to MM equation
- ✂ Bioproduct market USD 16 billion

3. Bioreactors II

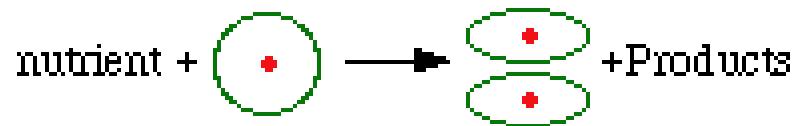
○ Bioconversion

- Advantages

- mild reaction conditions
- high yields
- enzyme acts as stereo-specific catalyst

- Bacteria as living chemical factory

- recombinant DNA



3. Bioreactors III

○ Cell growth

I. Lag phase

- adjusting new environment

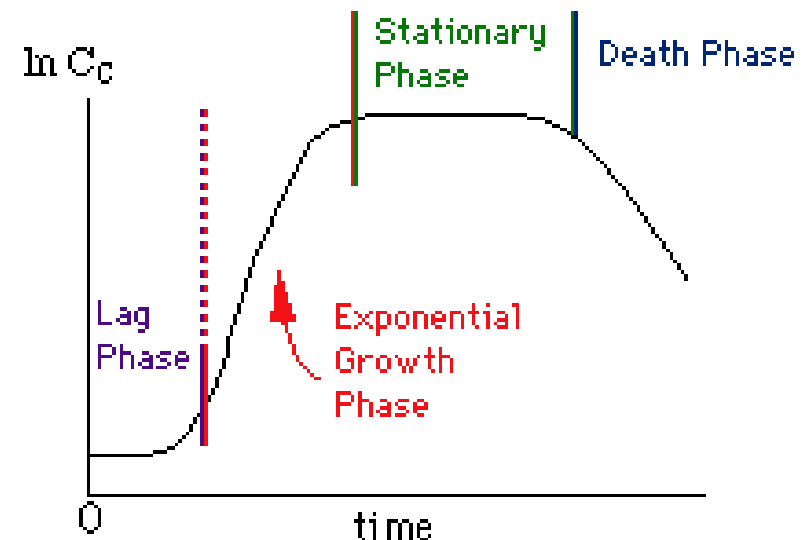
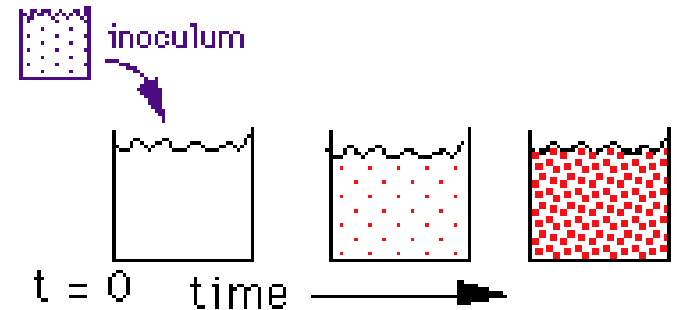
II. Exponential growth phase

- dividing at the maximum rate

III. Stationary phase

- cells reach minimum biological space
- **useful products**

IV. Death phase



3. Bioreactors IV

○ Rate Laws

Cells + Substrate → More cells + Product

- Monod equation

$$r_g = \mu C_C$$

$$\mu = \mu_{\max} \frac{C_S}{K_S + C_S}$$

$$r_g = \frac{\mu_{\max} C_S C_C}{K_S + C_S}$$

- In general $r_g = k_{OBS} \left(\frac{\mu_{\max} C_S}{K_S + C_S} \right) C_C$

$k_{OBS} = \left(1 - \frac{C_P}{C_P^*} \right)^n$ C_P^* = Where Product concentration at which all metabolism ceases.



3. Bioreactors V

- **Stoichiometry**

- **Yield coefficients**

$$Y_{C/S} = \frac{\text{Mass of new cells formed}}{\text{Mass of substrate consumed}} = -\frac{\Delta C_C}{\Delta C_S}$$

with $Y_{C/S} = 1/Y_{S/C}$

$$Y_{P/S} = \frac{\text{Mass of product formed}}{\text{Mass of substrate consumed}} = -\frac{\Delta C_P}{\Delta C_S}$$

- **Maintenance**

$$m = \frac{\text{Mass of substrate consumed for maintenance}}{\text{Mass of cells} \cdot \text{Time}}$$

3. Bioreactors VI

○ Stoichiometry 2

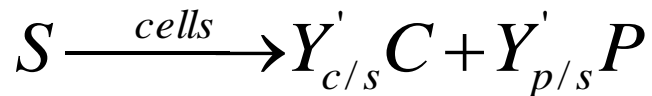
- Rate of substrate consumption for maintenance

$$r_{sm} = mC_c$$

- When maintenance can be neglected

$$C_c = Y_{c/s} [C_{s0} - C_s]$$

- Total substrate consumed is to form new cells (C) and product (P)



3. Bioreactors VII

- **Stoichiometry 3**

- **Substrate utilization**

- **Substrate balance**

- **during the growth phase \Rightarrow lumped into $Y_{s/c}$**

$$-r_s = Y_{s/c} r_g + mC_c$$

3. Bioreactors VIII

○ Stoichiometry 4

- Substrate balance 2

- corresponding rate of product formation

$$r_p = Y_{p/c} r_g$$

- for stationary phase, nutrient for growth virtually exhausted
- a different nutrient is used for cell maintenance
⇒ produce the desired product

$$r_p = \frac{k_p C_{sn} C_c}{K_{sn} + C_{sn}}$$

3. Bioreactors IX

○ Stoichiometry 5

- Substrate balance 3

- net rate of secondary nutrient consumption

$$\begin{aligned} -r_{sn} &= mC_c + Y_{sn/p} r_p \\ &= mC_c + \frac{Y_{sn/p} k_p C_{sn} C_c}{K_{sn} + C_{sn}} \end{aligned}$$

- the desired product can be obtained with no cell growth stage \Rightarrow concern about the change of secondary nutrient

$$C_p = Y_{p/s} [C_{sn0} - C_{sn}]$$

3. Bioreactors X

○ Stoichiometry 6

- Two limiting situations for substrate consumption to cell growth and product formation
 - product formed only during growth phase
 - product formed only during stationary phase

☞ Luedeking-Piret equation

$$q_p = \alpha \mu_g + \beta$$

→ growth (from α)

→ non-growth (from β)

with $r_p = q_p C_c$

3. Bioreactors XI

- **Mass Balance**

- **Counting the number of living cells**

- **Counting the mass of living cells**

3. Bioreactors XII

○ Mass Balance

- Batch operation

- **cell** $V \frac{dC_c}{dt} = r_g V - r_d V$ $\frac{dC_c}{dt} = r_g - r_d$

- **substrate** $V \frac{dC_c}{dt} = r_s V = Y_{s/c} (-r_g) V - m C_c V$

$$\frac{dC_c}{dt} = Y_{s/c} (-r_g) - m C_c$$

- **in stationary phase, no growth**

$$V \frac{dC_s}{dt} = -m C_c V + Y_{s/p} (-r_p) V$$

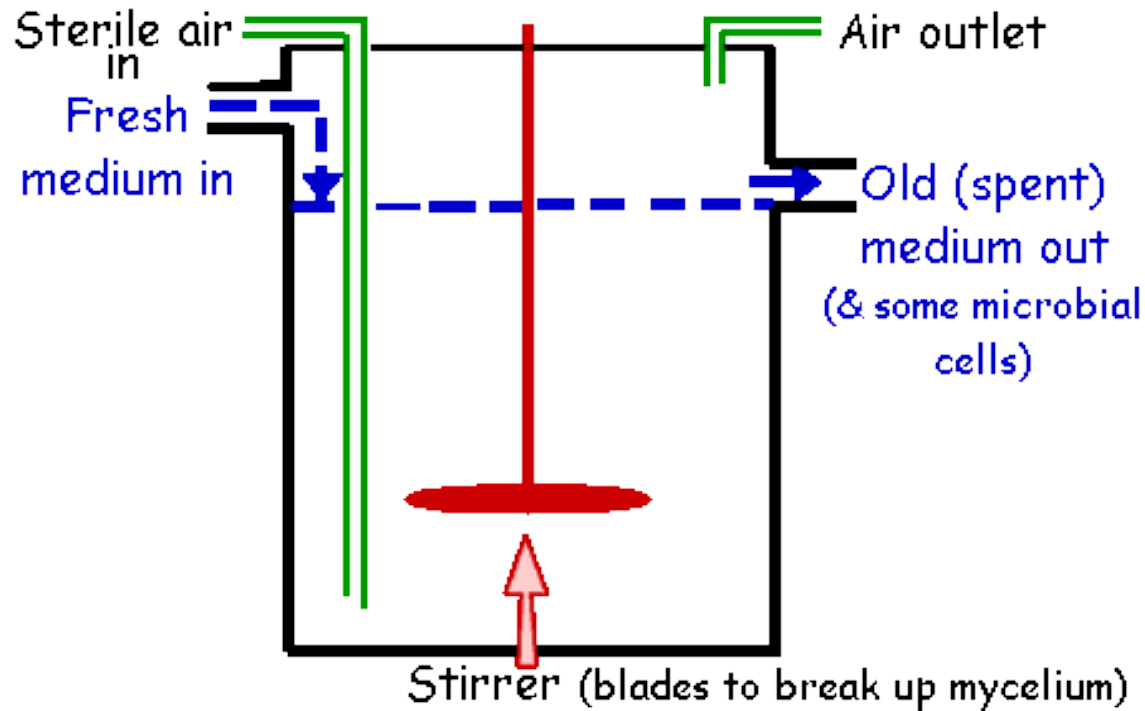
3. Bioreactors XIII

- Mass Balance 2
 - Batch operation
 - product

$$V \frac{dC_p}{dt} = r_p V = Y_{p/s} (-r_s) V$$

3. Bioreactors XIV

- Chemostats



3. Bioreactors XV

- Design Equations

- Dilution rate $D = \frac{v_0}{V}$

- Mass balance (CSTR)

- Monod equation

$$r_g = \mu C_c = \frac{\mu_{\max} C_s C_c}{K_s + C_s}$$

3. Bioreactors XVI

○ Design Equations 2

- Steady state operation

$$DC_c = r_g - r_d$$

$$D(C_{s0} - C_s) = -r_s$$

- Neglecting the death rate

$$F_c = C_c v_0 = r_g V = \mu C_c V$$

- Divide by $C_c V$ $D = \mu$

☞ specific growth rate of the cell can be controlled by varying D

3. Bioreactors XVII

○ Design Equations 3

- Steady state operation 2

- solving for steady state substrate conc.

$$C_c = \frac{DK_s}{\mu_{\max} - D}$$

- single nutrient limiting, substrate consumed to cell growth only, cell maintenance neglected

$$-r_s = r_g Y_{s/c} \quad C_c = Y_{c/s} (C_{s0} - C_s)$$

$$C_c = Y_{c/s} \left[C_{s0} - \frac{DK_s}{\mu_{\max} - D} \right]$$

3. Bioreactors XVIII

- Wash out

- Neglect death rate and cell maintenance
- Steady state

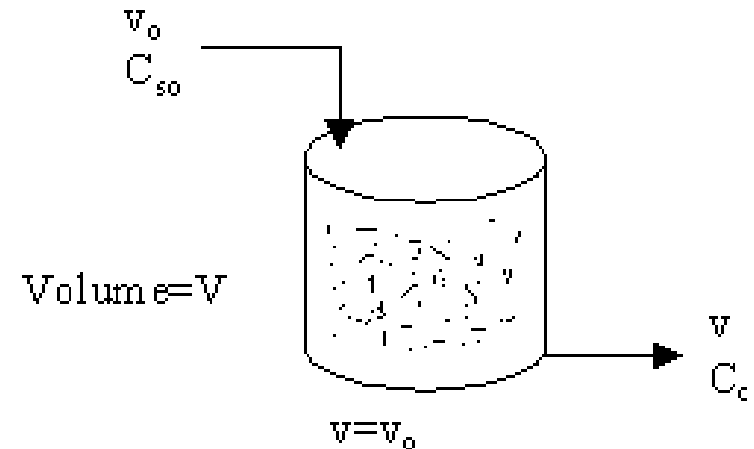
$$0 = -v C_C + r_g V$$

$$D C_C = r_g = \frac{\mu_{\max} C_S}{K_S + C_S} C_C = \mu C_C$$

$$D = \mu = \frac{\mu_{\max} C_S}{K_S + C_S}$$

$$C_S = \frac{D K_S}{\mu_{\max} - D}$$

$$C_C = Y_{C/S} [C_{S0} - C_S] = Y_{C/S} \left[C_{S0} - \frac{D K_S}{\mu_{\max} - D} \right]$$



3. Bioreactors XIX

○ Wash out 2 $D_{w} = \frac{\mu_{max} C_{s0}}{K_s + C_{s0}}$

- Maximum production rate

• dividing by the reactor volume, V , which is constant

production rate = $\dot{m}_c = v_o C_C$

$$\frac{\dot{m}}{V} = DC_C$$

• substituting for C_C

$$DC_C = DY_{c/s} \left(C_{s0} - \frac{DK_s}{\mu_{max} - D} \right)$$

