

# Chapter 14: Properties and Applications of Ceramics

## ISSUES TO ADDRESS...

- In what ways are ceramic phase diagrams different from phase diagrams for metals?
- How are the mechanical properties of ceramics measured, and how do they differ from those for metals?
- How do we classify ceramics?
- What are some applications of ceramics?

# Ceramic Phase Diagrams

MgO-Al<sub>2</sub>O<sub>3</sub> diagram:

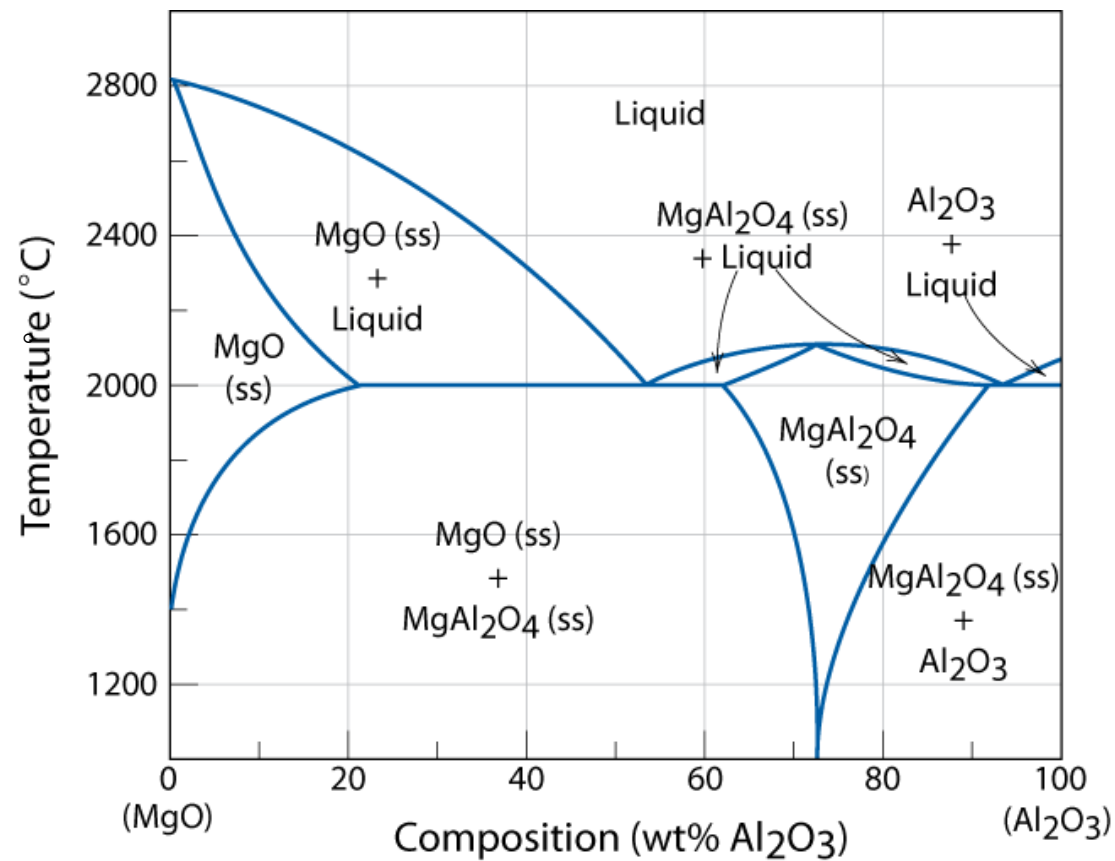


Fig. 14.2, Callister & Rethwisch 9e.

[Adapted from B. Hallstedt, "Thermodynamic Assessment of the System MgO-Al<sub>2</sub>O<sub>3</sub>," J. Am. Ceram. Soc., 75[6], 1502 (1992). Reprinted by permission of the American Ceramic Society.]

# Mechanical Properties

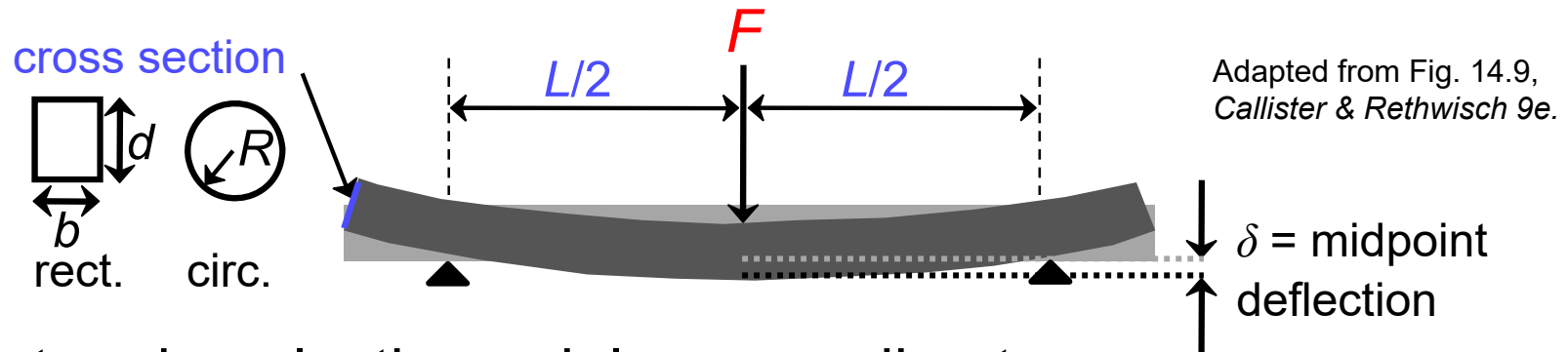
Ceramic materials are more brittle than metals.

Why is this so?

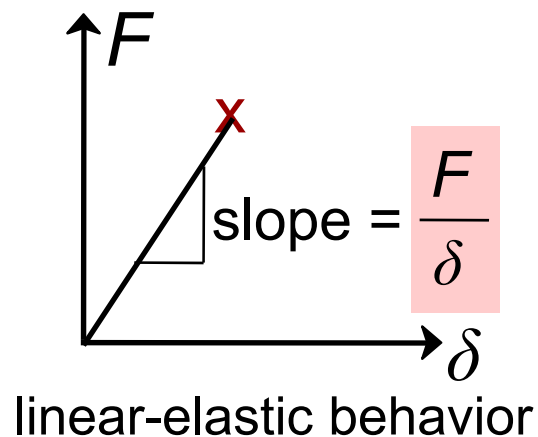
- Consider mechanism of deformation
  - In crystalline, by dislocation motion
  - In highly ionic solids, dislocation motion is difficult
    - few slip systems
    - resistance to motion of ions of like charge (e.g., anions) past one another

# Flexural Tests – Measurement of Elastic Modulus

- Room  $T$  behavior is usually elastic, with brittle failure.
- **3-Point Bend Testing** often used.
  - tensile tests are difficult for brittle materials.



- Determine elastic modulus according to:

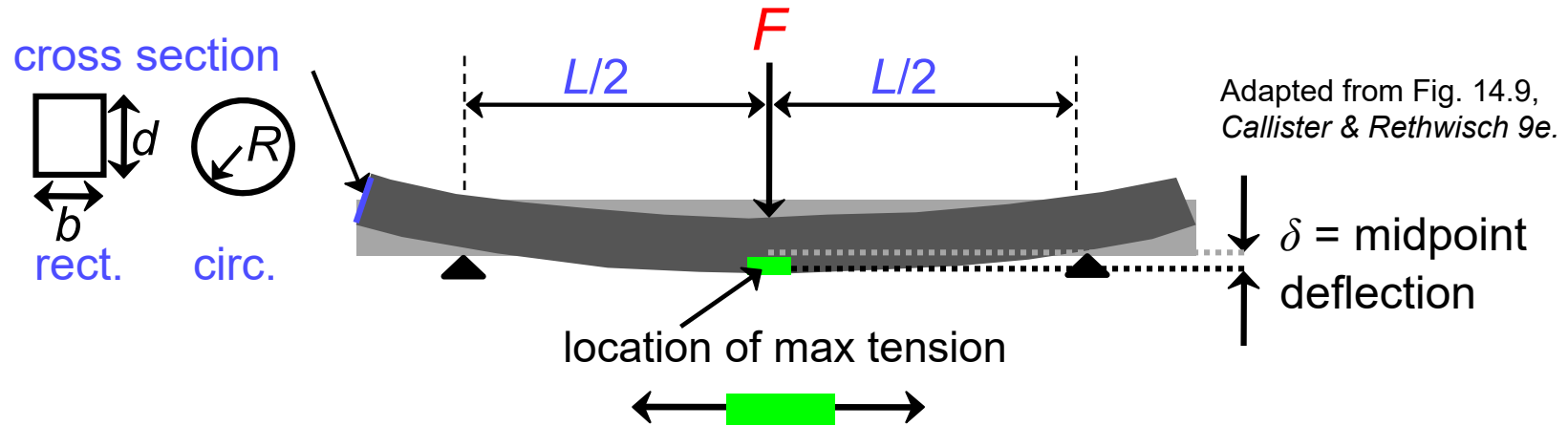


$$E = \frac{F}{\delta} \frac{L^3}{4bd^3} \quad (\text{rect. cross section})$$

$$E = \frac{F}{\delta} \frac{L^3}{12\pi R^4} \quad (\text{circ. cross section})$$

# Flexural Tests – Measurement of Flexural Strength

- 3-point bend test to measure room- $T$  flexural strength.



- Flexural strength:

$$\sigma_{fs} = \frac{3F_f L}{2bd^2} \quad (\text{rect. cross section})$$

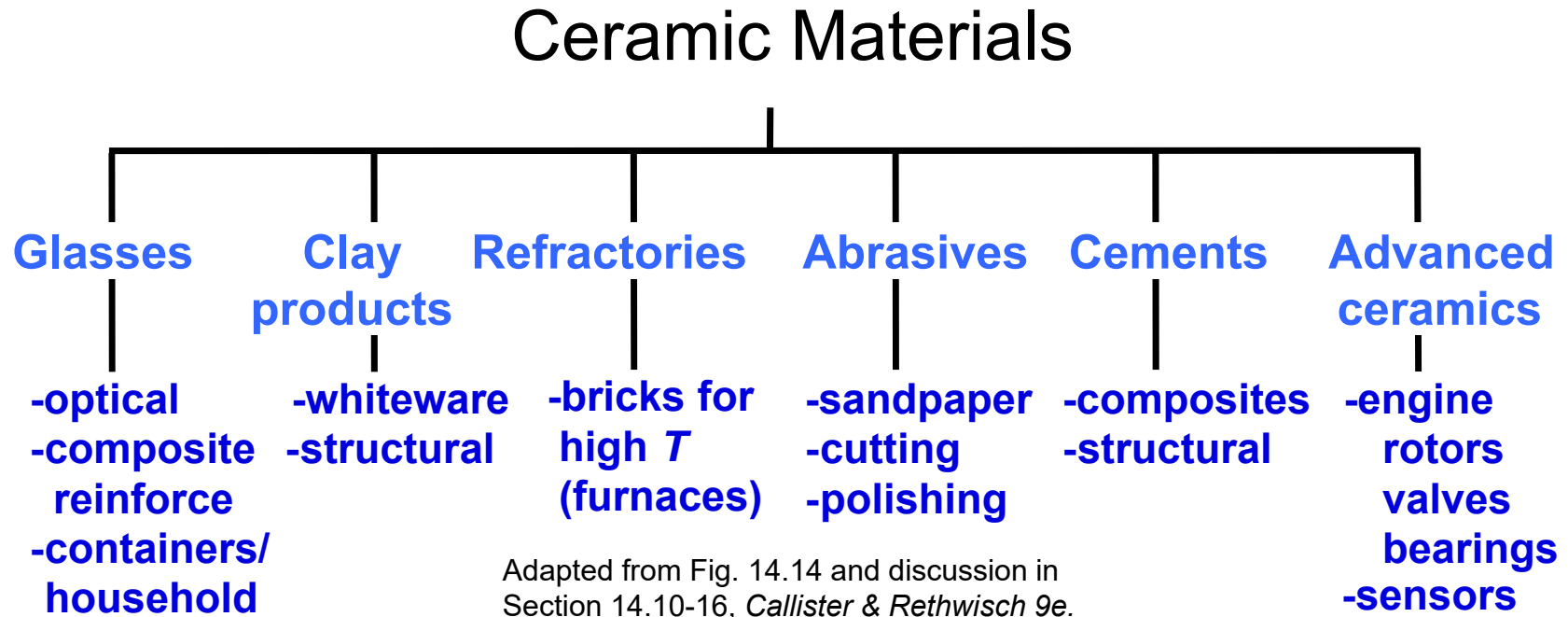
$$\sigma_{fs} = \frac{F_f L}{\pi R^3} \quad (\text{circ. cross section})$$

- Typical values:

Material	$\sigma_{fs}$ (MPa)	$E$ (GPa)
Si nitride	250-1000	304
Si carbide	100-820	345
Al oxide	275-700	393
glass (soda-lime)	69	69

Data from Table 14.1, Callister & Rethwisch 9e.

# Classification of Ceramics

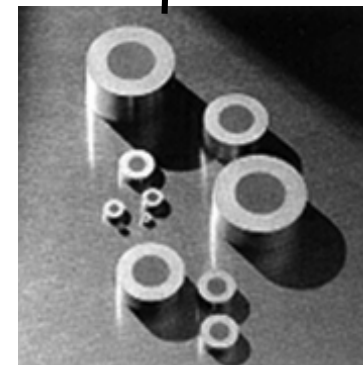


# Ceramics Application: Die Blanks

- Die blanks:
  - Need wear resistant properties!
- Die surface:
  - 4  $\mu\text{m}$  polycrystalline diamond particles that are sintered onto a cemented tungsten carbide substrate.
  - polycrystalline diamond gives uniform hardness in all directions to reduce wear.



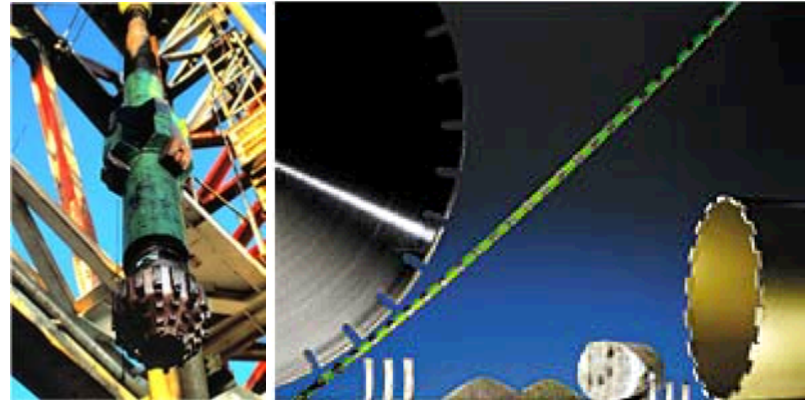
Adapted from Fig. 17.2(d),  
*Callister & Rethwisch 9e.*



Courtesy Martin Deakins, GE  
Superabrasives, Worthington,  
OH. Used with permission.

# Ceramics Application: Cutting Tools

- Tools:
  - for grinding glass, tungsten, carbide, ceramics
  - for cutting Si wafers
  - for oil drilling
- Materials:
  - manufactured single crystal or polycrystalline diamonds in a metal or resin matrix.
  - polycrystalline diamonds resharpen by microfracturing along cleavage planes.

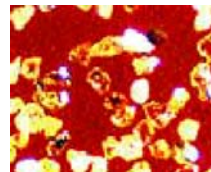


oil drill bits

blades



Single crystal diamonds



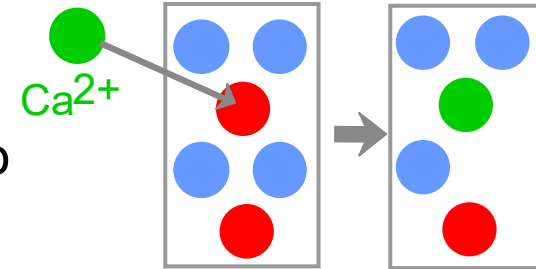
polycrystalline diamonds in a resin matrix.

Photos courtesy Martin Deakins, GE Superabrasives, Worthington, OH. Used with permission.



# Ceramics Application: Sensors

- Example:  $ZrO_2$  as an oxygen sensor
- Principle: Increase diffusion rate of oxygen to produce rapid response of sensor signal to change in oxygen concentration

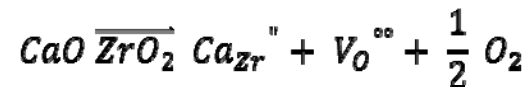


A substituting  $Ca^{2+}$  ion removes a  $Zr^{4+}$  ion and an  $O^{2-}$  ion.

- Approach:

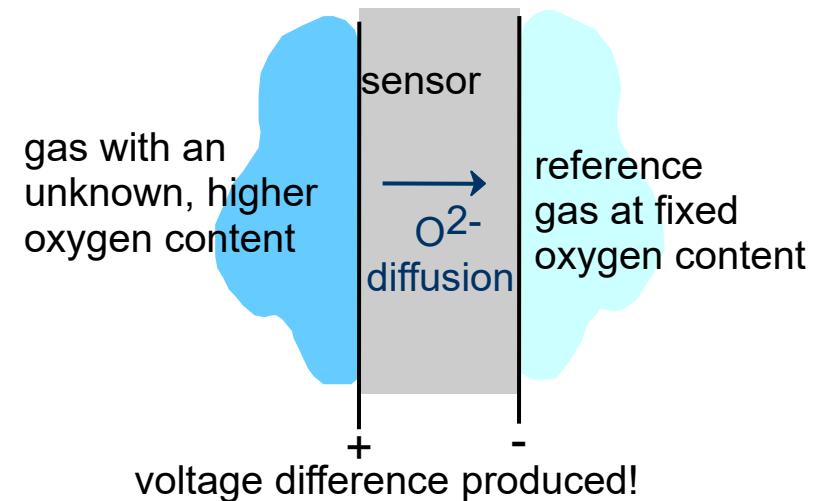
Add Ca impurity to  $ZrO_2$ :

- increases  $O^{2-}$  vacancies
- increases  $O^{2-}$  diffusion rate



- Operation:

- voltage difference produced when  $O^{2-}$  ions diffuse from the external surface through the sensor to the reference gas surface.
- magnitude of voltage difference  $\propto$  partial pressure of oxygen at the external surface



$$emf = \frac{RT}{4F} \ln \left( \frac{P'_{O_2}}{P''_{O_2}} \right)$$

# Refractories

- Materials to be used at high temperatures (e.g., in high temperature furnaces).
- Consider the Silica ( $\text{SiO}_2$ ) - Alumina ( $\text{Al}_2\text{O}_3$ ) system.
- Silica refractories - silica rich - small additions of alumina depress melting temperature (phase diagram):

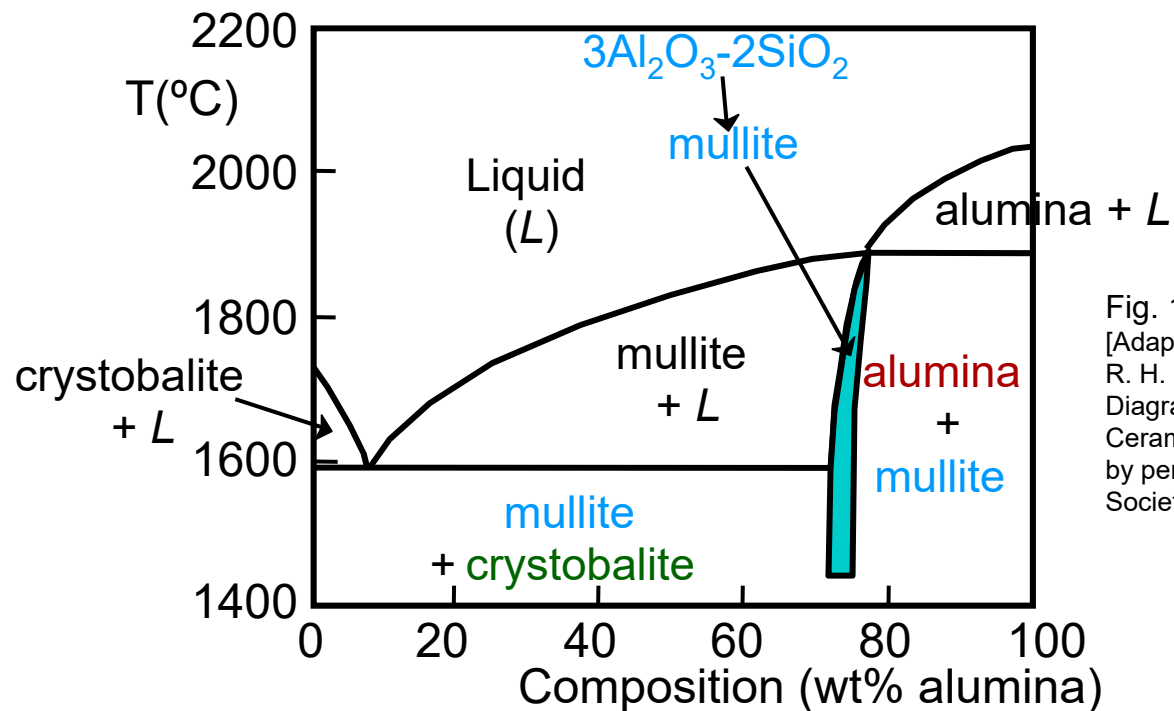


Fig. 14.4, *Callister & Rethwisch 9e*.  
[Adapted from F. J. Klug, S. Prochazka, and R. H. Doremus, "Alumina-Silica Phase Diagram in the Mullite Region," *J. Am. Ceram. Soc.*, 70[10], 758 (1987). Reprinted by permission of the American Ceramic Society.]

# Advanced Ceramics: Materials for Automobile Engines

- Advantages:
  - Operate at high temperatures – high efficiencies
  - Low frictional losses
  - Operate without a cooling system
  - Lower weights than current engines
- Disadvantages:
  - Ceramic materials are brittle
  - Difficult to remove internal voids (that weaken structures)
  - Ceramic parts are difficult to form and machine
- Potential candidate materials:  $\text{Si}_3\text{N}_4$ ,  $\text{SiC}$ , &  $\text{ZrO}_2$
- Possible engine parts: engine block & piston coatings

# Advanced Ceramics: Materials for Ceramic Armor

## Components:

- Outer facing plates
- Backing sheet

## Properties/Materials:

- Facing plates -- hard and brittle
  - fracture high-velocity projectile
  - $\text{Al}_2\text{O}_3$ ,  $\text{B}_4\text{C}$ ,  $\text{SiC}$ ,  $\text{TiB}_2$
- Backing sheets -- soft and ductile
  - deform and absorb remaining energy
  - aluminum, synthetic fiber laminates

# Nanocarbons

- **Fullerenes** – spherical cluster of 60 carbon atoms,  $C_{60}$ 
  - Like a soccer ball
- **Carbon nanotubes** – sheet of graphite rolled into a tube
  - Ends capped with fullerene hemispheres

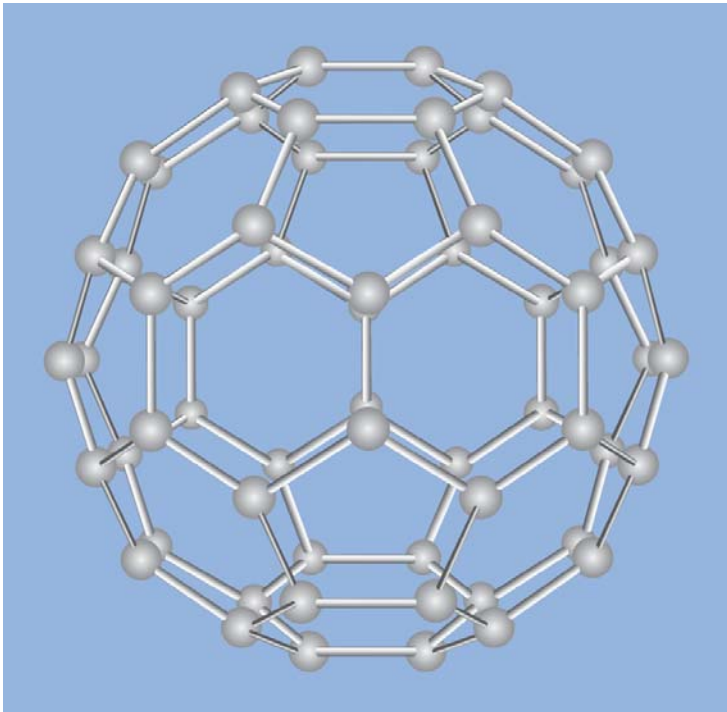


Fig. 14.20, *Callister & Rethwisch 9e.*

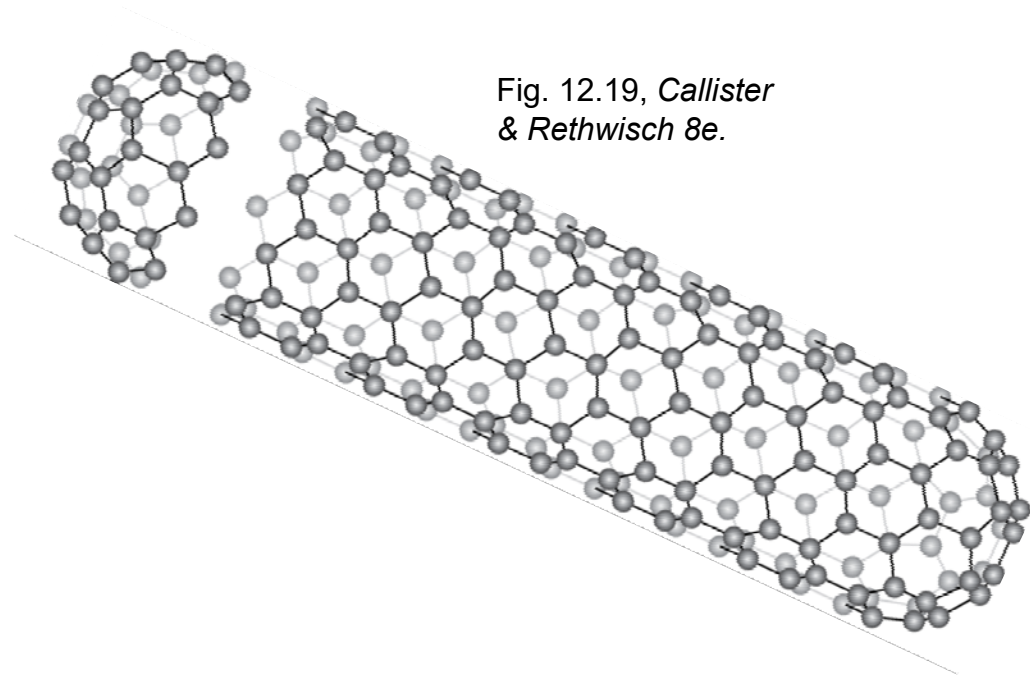


Fig. 12.19, *Callister & Rethwisch 8e.*

## Nanocarbons (cont.)

- **Graphene** – single-atomic-layer of graphite
  - composed of hexagonally  $sp^2$  bonded carbon atoms

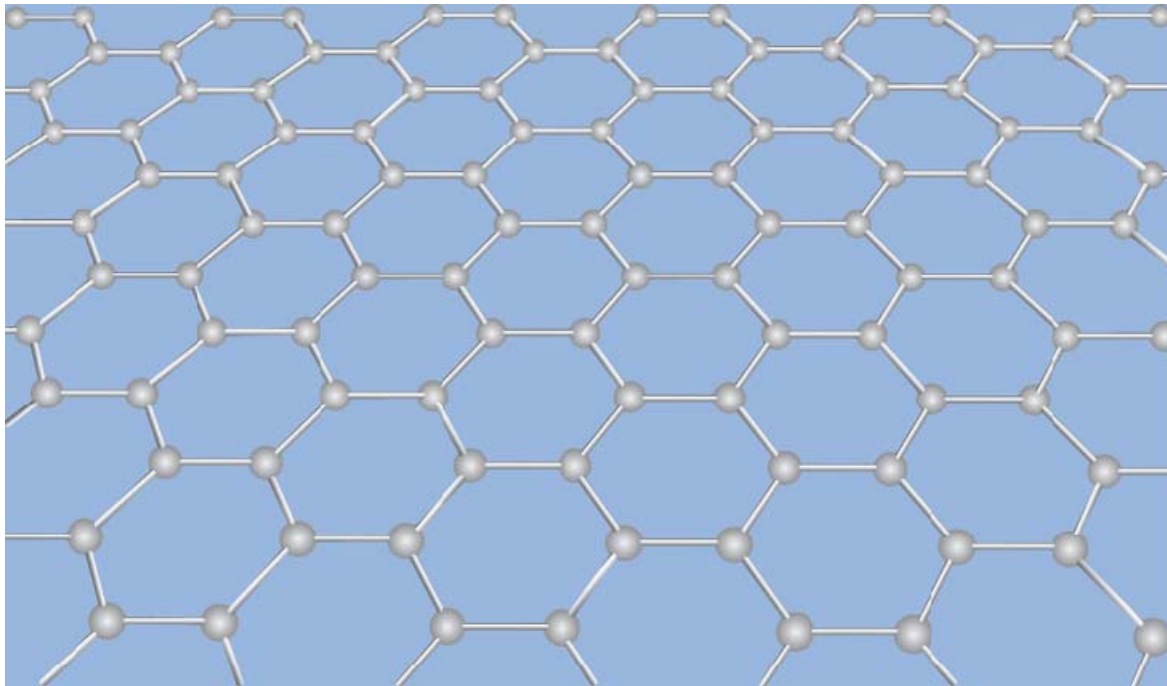
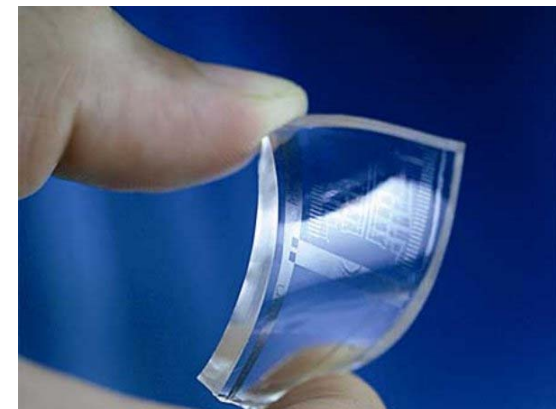
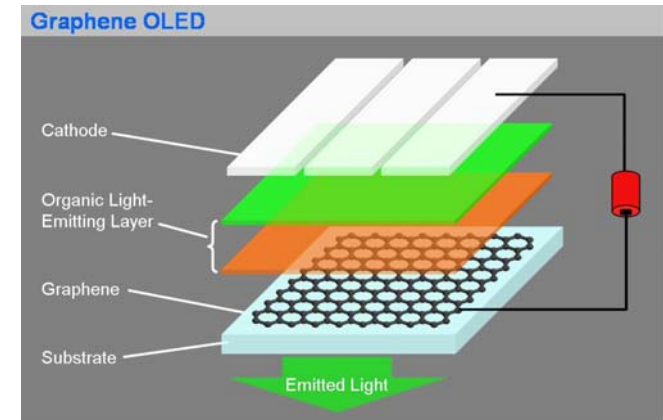
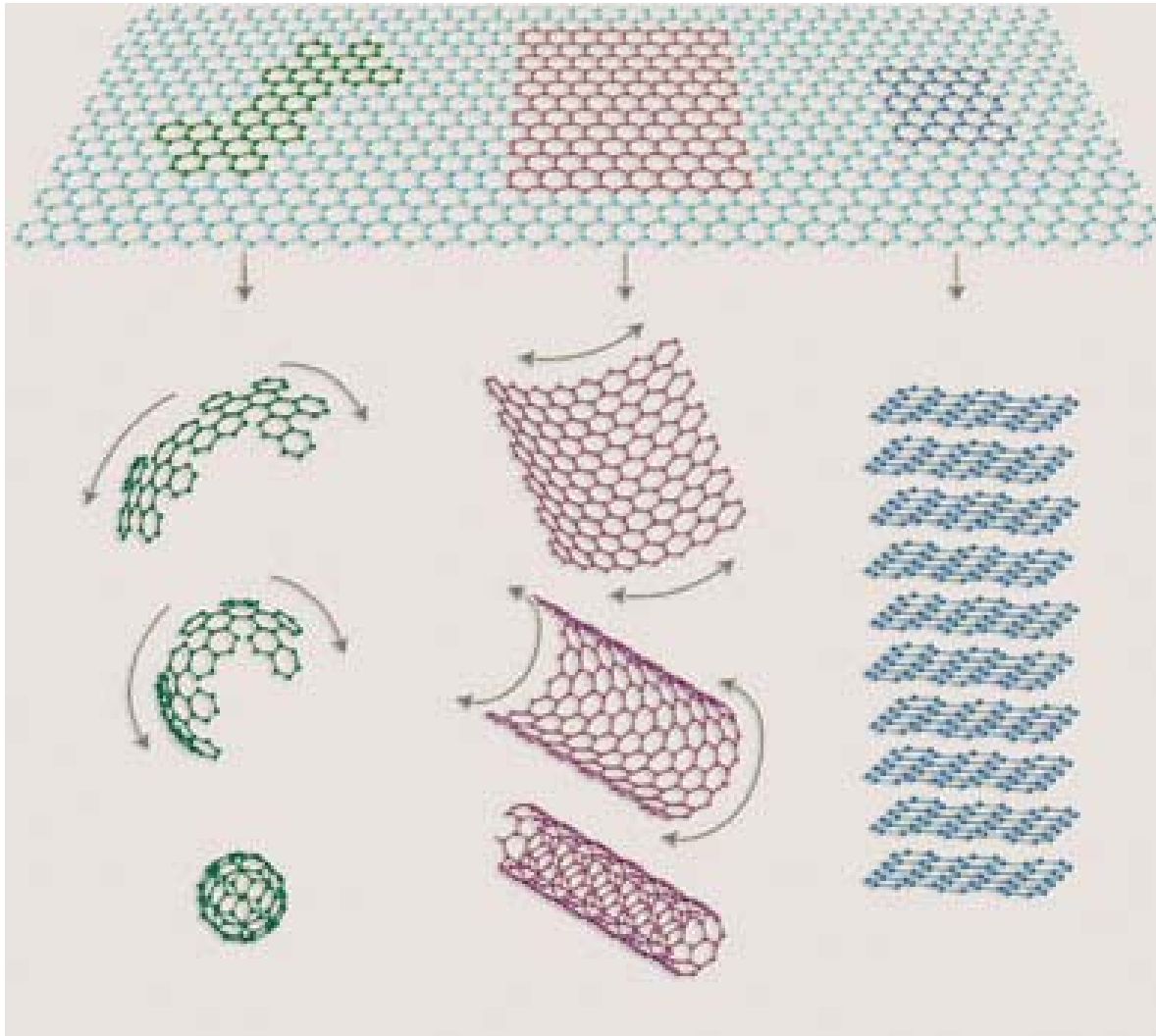
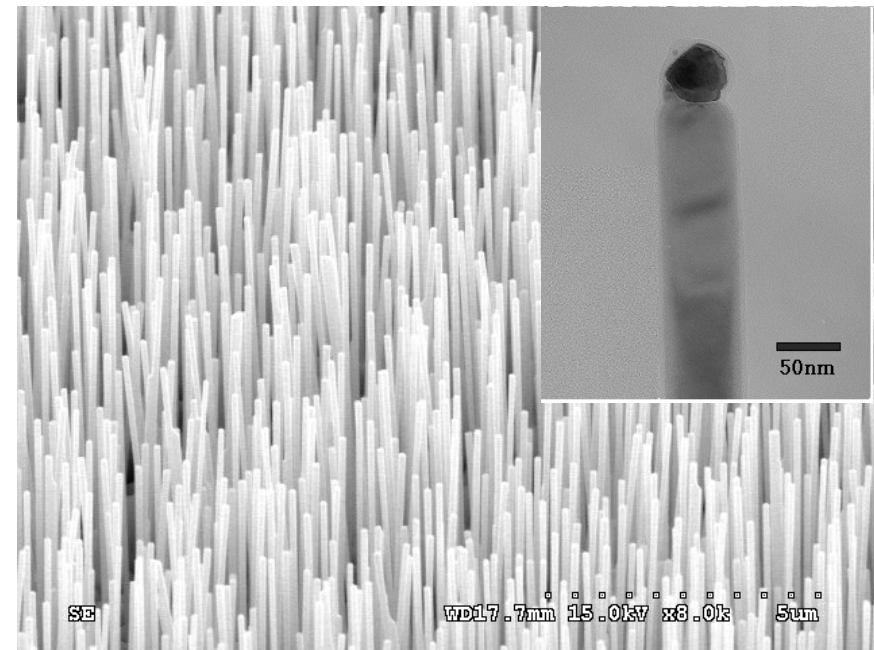
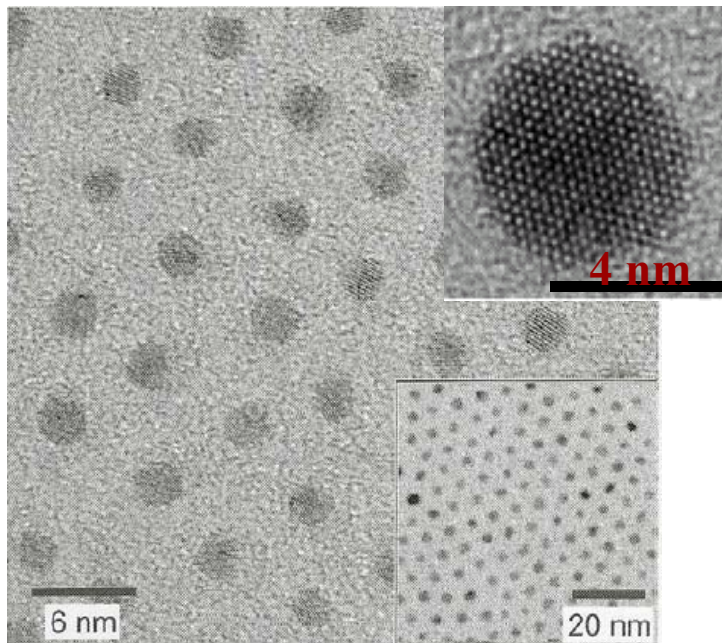
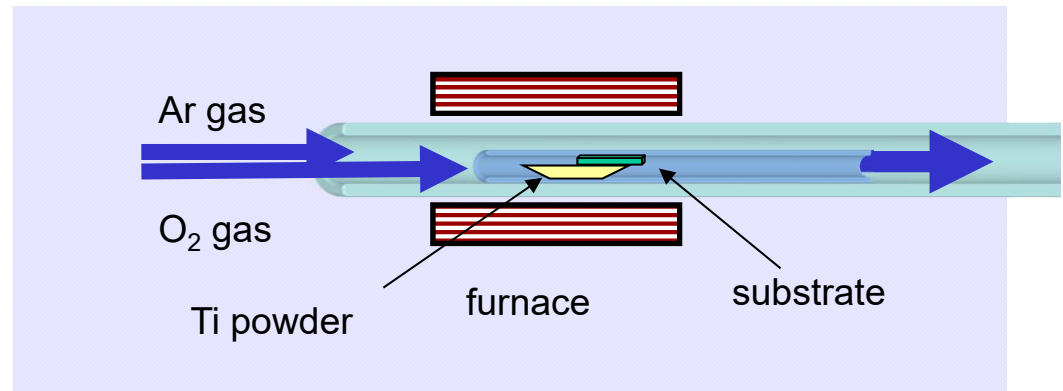


Fig. 14.22, Callister & Rethwisch 9e.

- **Graphenes** – Sheet of graphite



- Synthesis of Nanomaterials



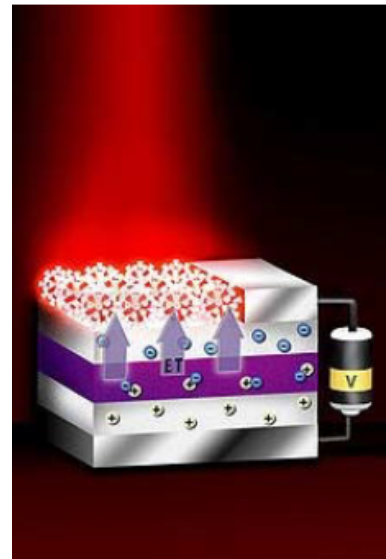


# Applications of Nanomaterials (0-D)

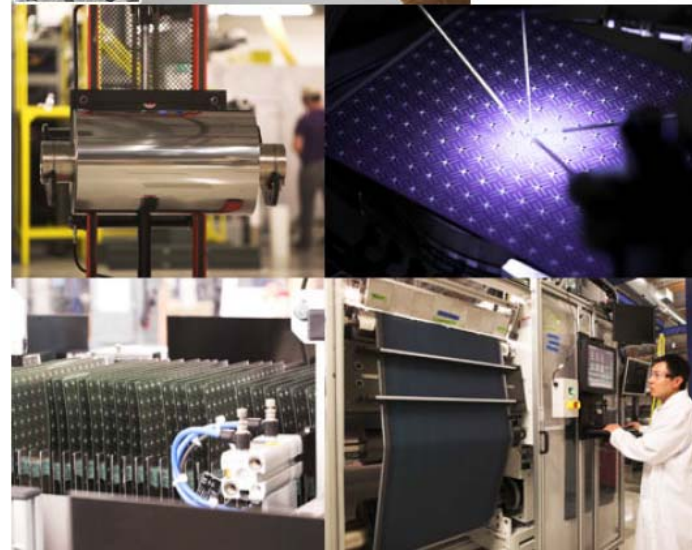
## QD Biosensing



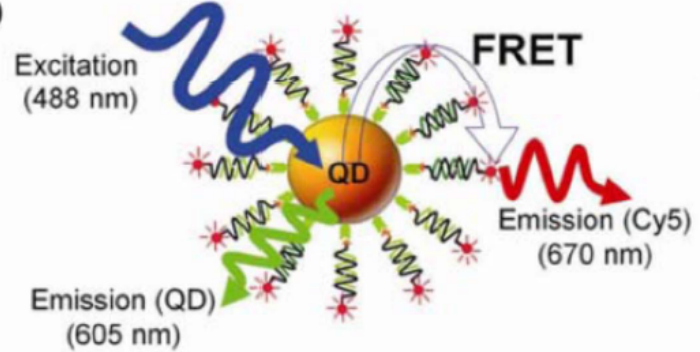
## QD LED



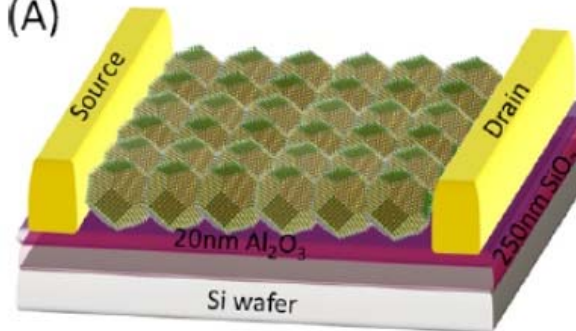
## QD solar cells



## Fluorescence Resonance Energy Transfer



(A)

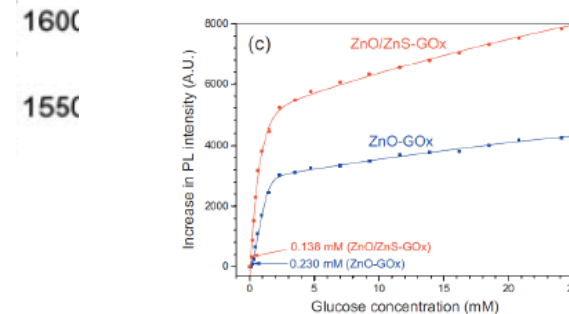
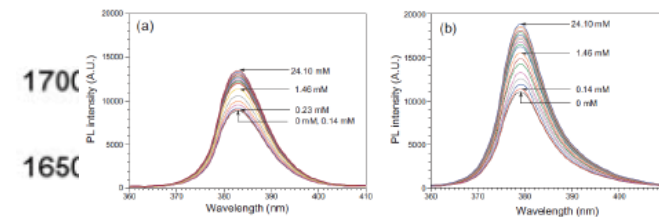
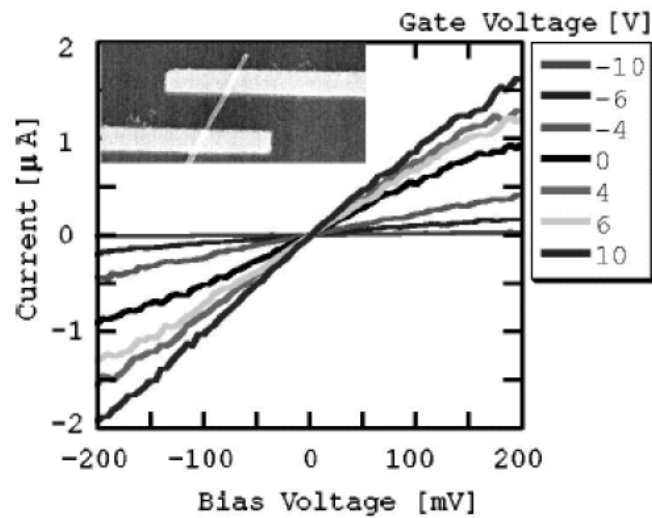
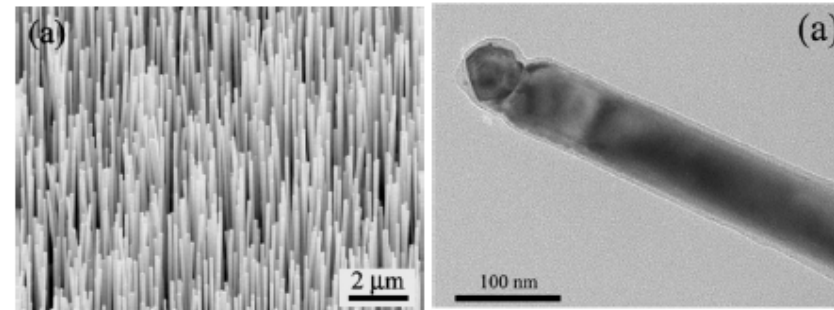
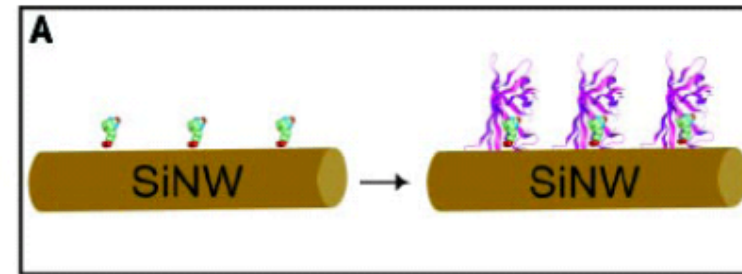
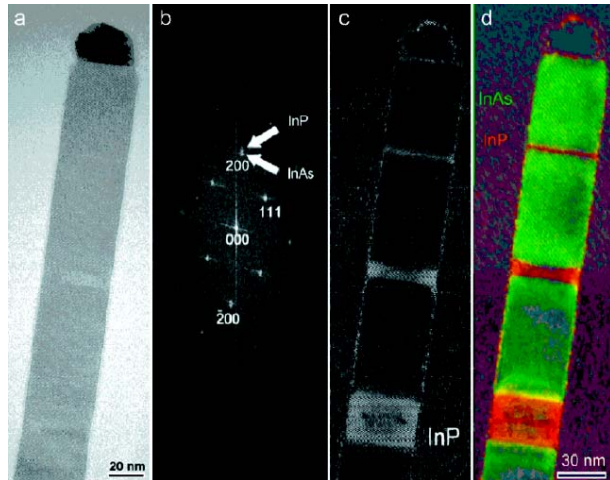


## QD Transistor

# Applications of Nanomaterials (1-D)

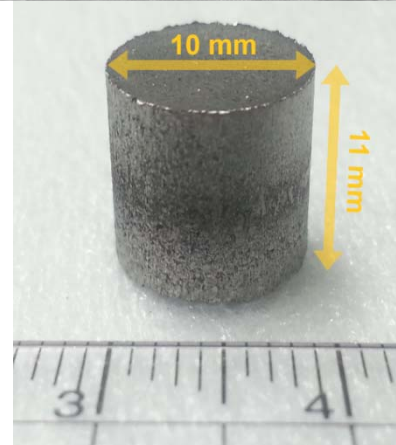
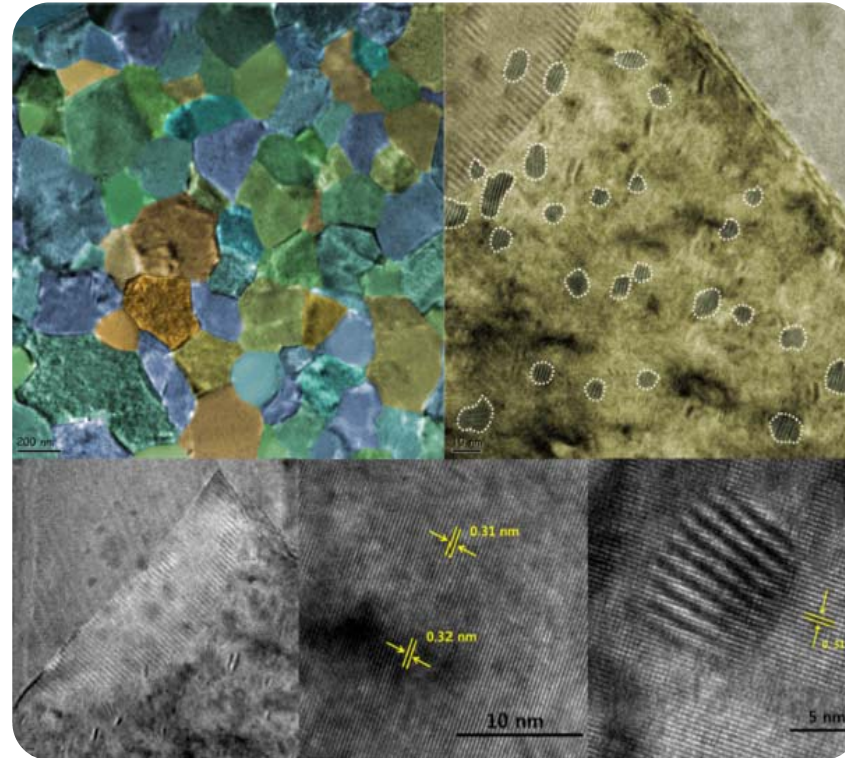
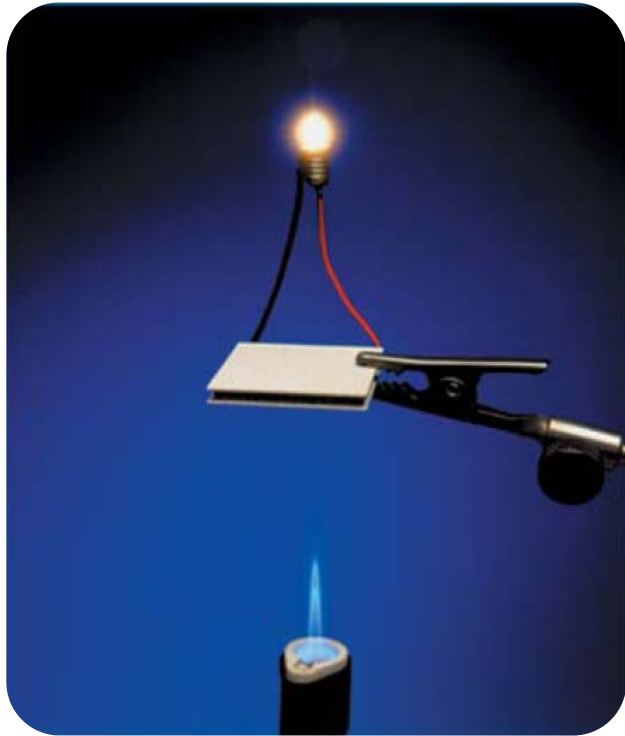
## NW Biosensor

### NW Transistor





# Applications of Nanomaterials (3-D)



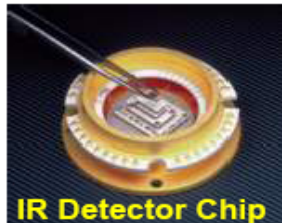
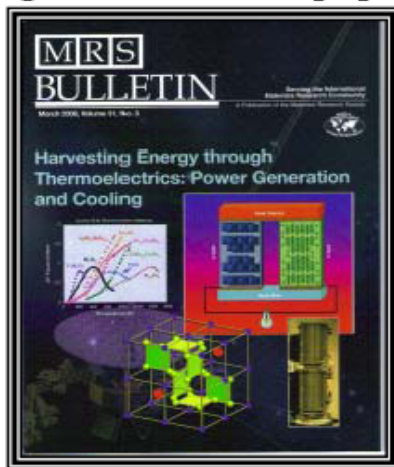


# Today's TE Applications

## Cassini - Saturn - NASA



40 years of NASA Investment in High Temperature TE Power Generation Technology for Deep Space Science Exploration



IR Sensing



All-Solid-State Refrigeration



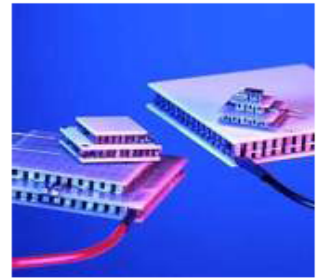
TE Powered Watch



Waste Heat Recovery



Autonomous Remote Power Generator



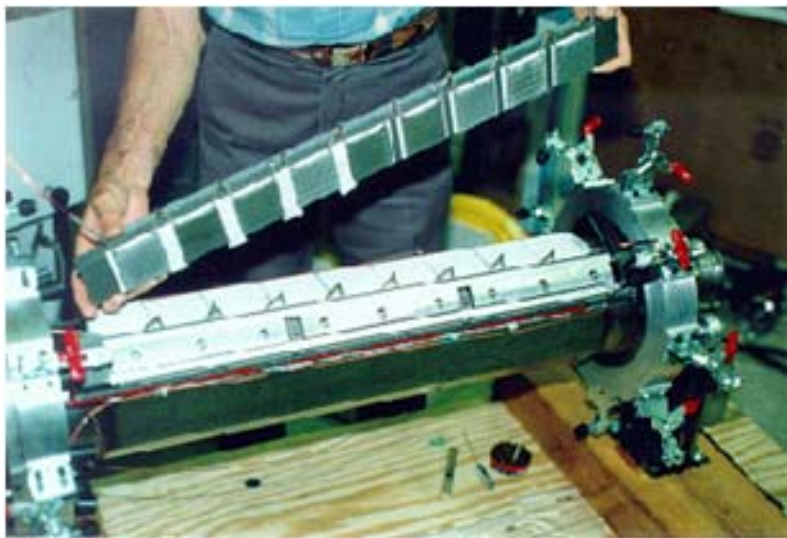
TE Module A Key Enabling Technology Component



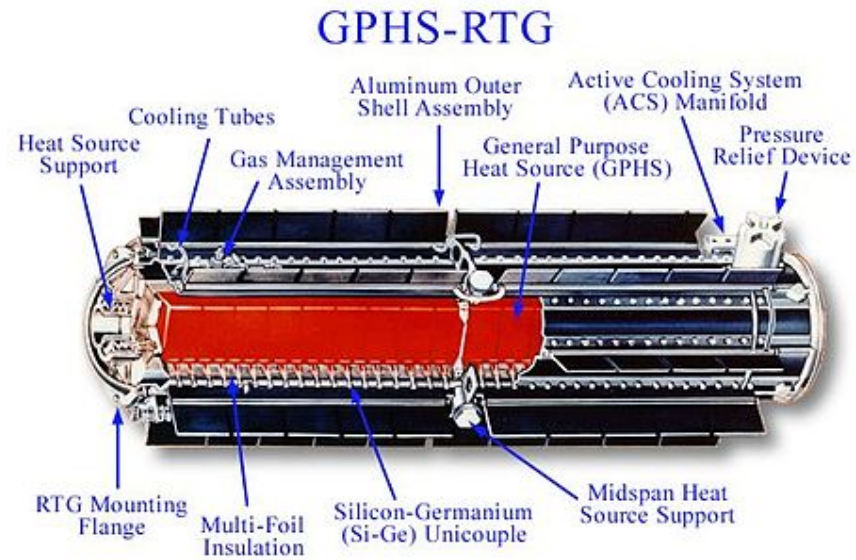
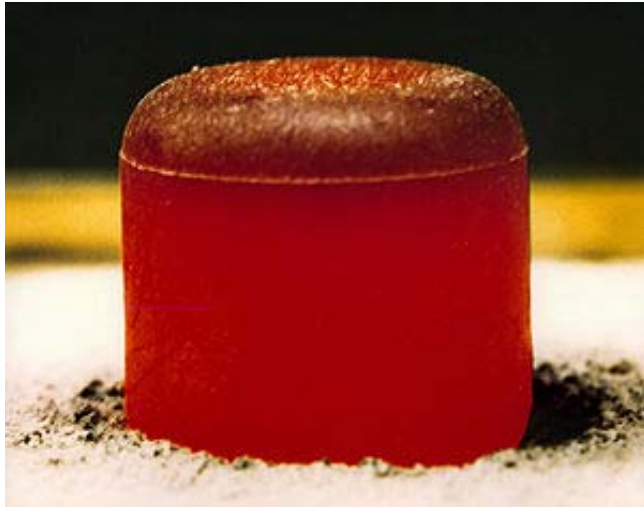
Active heated and cooled seating system



# TE Power Generation (TEG) for Hybrid Cars



# Radioisotope thermoelectric generator



# Summary

- Room-temperature mechanical behavior – flexural tests
  - linear-elastic; measurement of elastic modulus
  - brittle fracture; measurement of flexural modulus
- Categories of ceramics:
  - glasses
  - refractories
  - advanced ceramics
  - clay products
  - cements